

CHEMISTRY UNIT 1 STUDY GUIDE.

What is Matter?

- Solid, Liquid, gas
- Anything that takes up space & has mass
- Mass is the amount of matter in an object.

Types

- Substance- a particular kind of matter — pure - made up of elements and compounds
- Mixture- more than one kind of matter. 2+ elements and compounds

Properties:

Physical – property that can be observed and measured without changing the substance.

Chemical – property that can only be observed by changing the type of substance.

Extensive- only depends on the amount of matter

Mass, weight, volume, length

Intensive – only depends on the type of matter, not the amount used to identify a substance.

Color, Odor, melting/ freezing point.

Ductility- the ability to make copper into wires.

States of Matter:

	Definite volume	Definite shape	Temperature increase	compressible
SOLID	Yes	Yes	Small expans	No
LIQUID	Yes	No Takes up shape of container	Small expans	No
GAS	No	No	Large expans	Yes



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Environment, Nuclear Safety and Civil Protection,1993 **The McGuffey Reader** ,1937 Science Education in Canadian

Schools. Volume III. Case Studies of Science Teaching. Background Study 52 John Olson,Science Council of Canada, Ottawa
(Ontario),,1984 Volume 3 of a three volume study of science education in Canadian schools reporting 8 case studies of

science teaching in action in Canadian schools Sites for intensive study were selected to include a diversity of both regions
and school settings and at each school site various kinds of information were collected to obtain as complete a picture as

possible Major issues addressed include integration and options as forms of curriculum organization socialization as a
priority the inquiry approach and understanding and the dynamics of change and dilemmas of practice Chemistry and
Industry Review ,1928 Graduate Student Support and Manpower Resources in Graduate Science Education National

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