

Chemical Reactions:

1. The following reaction $2\text{H}_2\text{O}_2 \rightarrow 2\text{H}_2\text{O} + \text{O}_2$ is given. What is the ΔH°_f for the reaction?
- A) greater than 2.00 kcal/mol
 B) greater than 1.00 kcal/mol
 C) less than 1.00 kcal/mol
 D) less than 2.00 kcal/mol
 E) greater than 4.00 kcal/mol
2. A 1.0-mol sample of PCl_3 reacts with excess oxygen according to the equation: $2\text{PCl}_3(\text{g}) \rightarrow 2\text{PCl}_5(\text{g}) + \text{O}_2(\text{g})$. Below, which reaction will yield the greatest amount of O_2 after PCl_3 is removed?
- A) 0.0 mol
 B) 1.0 mol
 C) 1.2 mol
 D) 1.7 mol
 E) 2.0 mol
3. The reaction of NO_2 is step characterized by the equation: $4\text{NO}_2 \rightarrow 3\text{O}_2 + 2\text{N}_2\text{O}_5$. If 1.0 mol of NO_2 is added to 1.0 mol of N_2O_5 and the reaction is allowed to proceed, how will the reaction rate change?
- A) 1.000 mol/l
 B) 0.750 mol/l
 C) 1.000 mol/l
 D) 1.50 mol/l
 E) 2.00 mol/l
4. Calculated are the following reactions, or from the experiments, are unknowns, calculated.
- $\text{H}_2\text{O}_2(\text{l}) + \text{H}_2\text{O}_2(\text{l}) \rightarrow 2\text{H}_2\text{O}_2(\text{l}) = 200\text{O}_2(\text{g})$
- It is 7.5×10^{-3} mol of O_2 released for each mole of H_2O_2 . What is the calculated ΔH°_f ?
- A) -600 kJ/mol
 B) -300 kJ/mol
 C) 0 kJ/mol
 D) 300 kJ/mol
 E) 600 kJ/mol
5. Predict the following reaction:
- $\text{NO}_2(\text{g}) + \text{NO}_2(\text{g}) \rightarrow \text{NO}_2(\text{g}) + \text{NO}_2(\text{g}) +$
 $\text{NO}_2(\text{g})$
- Major steady-state levels of NO_2 are given below. In which below is the ΔH°_f calculated to be closest to zero?
- A) 1.5×10^{-3}
 B) 2.0×10^{-3}
 C) 1.0×10^{-3}
 D) 1.0×10^{-4}
 E) Insufficient information

6. The following reaction $2\text{H}_2(\text{g}) + \text{O}_2(\text{g}) \rightarrow 2\text{H}_2\text{O}(\text{l})$ is performed. Below, which number of reactions would the greatest amount of 1.00 mol of hydrogen and oxygen be required?

- A) 1.00 mol
 B) 1.50 mol
 C) 1.75 mol
 D) 2.00 mol
 E) 2.50 mol

7. Predict the following reaction for the following reaction:

$\text{C}_2\text{H}_6 + \text{O}_2 \rightarrow \text{CH}_3\text{OH} + \text{H}_2\text{O}$

What mass of oxygen, O_2 , is required to react completely with 1.0×10^{-3} g of C_2H_6 ?

- A) 2.0×10^{-3} g
 B) 2.5×10^{-3} g
 C) 4.0×10^{-3} g
 D) 4.5×10^{-3} g
 E) 5.0×10^{-3} g

8. Which reaction minimization and maximization reaction will be calculated for each of the following reactions?



- A) 1.00×10^{-3} mol/l reaction
 B) 1.00×10^{-3} mol/l reaction
 C) 1.50×10^{-3} mol/l reaction
 D) 2.00×10^{-3} mol/l reaction
 E) 2.50×10^{-3} mol/l reaction

9. Calculate the mass of carbon dioxide produced from 1.0×10^{-3} g of octane, C_8H_{18} , in the following reaction:



- A) 7.5×10^{-3} g CO_2
 B) 1.5×10^{-3} g CO_2
 C) 2.0×10^{-3} g CO_2
 D) 1.20×10^{-3} g CO_2
 E) 1.80×10^{-3} g CO_2

10. Calculate the mass of oxygen required to react with 1.0×10^{-3} g of sulfur in the following reaction:



- A) 1.00×10^{-3} mol/l reaction O_2
 B) 1.00×10^{-3} mol/l reaction O_2
 C) 1.75×10^{-3} mol/l reaction O_2
 D) 2.00×10^{-3} mol/l reaction O_2
 E) 2.50×10^{-3} mol/l reaction O_2

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Journal of Chemical Education, 1926 Includes Report of New England Association of Chemistry Teachers and Proceedings of the Pacific Southwest Association of Chemistry Teachers *Tests* Richard C. Sweetland, Daniel J. Keyser, 1983 Descriptions of over 3000 tests in English intended as a guide for psychologists educators and other personnel who need test information to meet their assessment needs Topical arrangement under 3 main sections of psychology education and business Each entry gives test name associated personal names intended age group purpose description time range scoring cost and publisher Indexes by test titles authors publishers visually impaired tests and scoring services

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Computer-assisted Instruction in Chemistry: General approach James S. Mattson,Harry B. Mark,Hubert C. MacDonald,1974 **Proceedings of Conference on Computers in the Undergraduate Curricula ,1973**

Media Review ,1986 **Nigerian Journal of Research and Production ,2010-11** **Science Instruction in the Middle and Secondary Schools** Eugene L. Chiappetta,Thomas R. Koballa (Jr.),Alfred T. Collette,1998

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