

**Chemical Reactions:**

1. The following reaction  $2\text{H}_2\text{O}_2 \rightarrow 2\text{H}_2\text{O} + \text{O}_2$  is given. What is the  $\Delta H^\circ_f$  for the reaction?

A) greater than 2.00 kcal/mol  
 B) greater than 1.00 kcal/mol  
 C) less than 1.00 kcal/mol  
 D) less than 2.00 kcal/mol  
 E) greater than 4.00 kcal/mol

2. A 1.0-mol sample of  $\text{PCl}_3$  reacts with excess oxygen according to the equation:  $2\text{PCl}_3(\text{g}) \rightarrow 2\text{PCl}_5(\text{g}) + \text{O}_2(\text{g})$ . Below, which reaction will yield the greatest amount of  $\text{O}_2$  after  $\text{PCl}_3$  is removed?

A) 0.0 mol  
 B) 1.0 mol  
 C) 1.2 mol  
 D) 1.7 mol  
 E) 2.0 mol

3. The reaction of  $\text{NO}_2$  is step characterized by the equation:  $4\text{NO}_2 \rightarrow 3\text{O}_2 + 2\text{N}_2\text{O}_5$ . If 1.0 mol of  $\text{NO}_2$  is added to 1.0 mol of  $\text{N}_2\text{O}_5$  and the reaction is allowed to proceed, how will the reaction rate change?

A) 1.000 mol/l  
 B) 0.750 mol/l  
 C) 1.000 mol/l  
 D) 1.50 mol/l  
 E) 2.00 mol/l

4. Calculated are the following reactions, or from the experiments, are unknowns, calculated.

$\text{H}_2\text{O}_2(\text{l}) + \text{H}_2\text{O}_2(\text{l}) \rightarrow 2\text{H}_2\text{O}_2(\text{l}) = 200\text{O}_2(\text{g})$

It is  $7.5 \times 10^{-3}$  mol of  $\text{O}_2$  released during the first 10 s of this reaction. The reaction rate is 0.5 mol of  $\text{H}_2\text{O}_2$  released in the first 10 s.

A)  $\text{O}_2$   
 B)  $\text{H}_2$   
 C)  $\text{H}_2$   
 D)  $\text{O}_2$   
 E)  $\text{C}$

5. Refer to the following reaction:

$\text{NO}_2(\text{g}) + \text{NO}_2(\text{g}) \rightarrow 4\text{NO}(\text{g}) + \text{O}_2(\text{g})$

Major steady-state levels of  $\text{NO}_2$  are given below. In which below is the  $\Delta H^\circ_f$  most likely to be negative?

A) 1.5 kcal  
 B)  $1.50 \times 10^3$  kcal  
 C)  $1.50 \times 10^2$  kcal  
 D) 1.50  
 E) Insufficient information

6. The following reaction  $2\text{H}_2\text{O}_2 \rightarrow 2\text{H}_2\text{O} + \text{O}_2$  is given. Below, which reaction will yield the greatest amount of  $\text{O}_2$  after 1.00 mol of hydrogen and oxygen react?

A) 1.00 mol  
 B) 1.20 mol  
 C) 1.500 mol  
 D) 2.00 mol  
 E) 2.500 mol

7. Refer to the following reaction:

$\text{C}_2\text{H}_6 + \text{O}_2 \rightarrow \text{CH}_3\text{OH} + \text{H}_2\text{O}$

What mass of oxygen,  $\text{O}_2$ , is required to react completely with  $1.0 \times 10^{-3}$  g of  $\text{C}_2\text{H}_6$ ?

A)  $2.00 \times 10^{-3}$  g  
 B)  $2.00 \times 10^{-2}$  g  
 C)  $4.00 \times 10^{-3}$  g  
 D)  $4.00 \times 10^{-2}$  g  
 E)  $0.100 \times 10^{-3}$  g

8. Refer reaction neutralization and synthesis of ammonia. In the following reaction, which is the limiting reactant?



A)  $1.00 \times 10^{-3}$  mol carbon  
 B)  $1.00 \times 10^{-3}$  mol carbon  
 C)  $4.00 \times 10^{-3}$  mol carbon  
 D)  $8.00 \times 10^{-3}$  mol carbon  
 E)  $0.100 \times 10^{-3}$  mol carbon

9. Calculate the mass of carbon dioxide produced from  $1.0 \times 10^{-3}$  g of carbon.  $\text{C}_2\text{H}_6 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$



A)  $7.0 \times 10^{-4}$  g  $\text{CO}_2$   
 B)  $1.00 \times 10^{-3}$  g  $\text{CO}_2$   
 C)  $2.00 \times 10^{-3}$  g  $\text{CO}_2$   
 D)  $1.000 \times 10^{-3}$  g  $\text{CO}_2$   
 E)  $1.000 \times 10^{-2}$  g  $\text{CO}_2$

10. Calculate the mass of oxygen required to react with  $1.0 \times 10^{-3}$  g of sulfur in the following reaction:



A)  $1.00 \times 10^{-3}$  mol carbon  $\text{O}_2$   
 B)  $0.250 \times 10^{-3}$  mol carbon  $\text{O}_2$   
 C)  $1.75 \times 10^{-3}$  mol carbon  $\text{O}_2$   
 D)  $2.00 \times 10^{-3}$  mol carbon  $\text{O}_2$   
 E)  $0.500 \times 10^{-3}$  mol carbon  $\text{O}_2$

# Multiple Choice Stoichiometry Test

**McGraw-Hill**

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**Journal of Chemical Education**, 1926 Includes Report of New England Association of Chemistry Teachers and Proceedings of the Pacific Southwest Association of Chemistry Teachers *Tests* Richard C. Sweetland, Daniel J. Keyser, 1983 Descriptions of over 3000 tests in English intended as a guide for psychologists educators and other personnel who need test information to meet their assessment needs Topical arrangement under 3 main sections of psychology education and business Each entry gives test name associated personal names intended age group purpose description time range scoring cost and publisher Indexes by test titles authors publishers visually impaired tests and scoring services

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**Media Review ,1986** **Nigerian Journal of Research and Production ,2010-11** **Science Instruction in the Middle and Secondary Schools** Eugene L. Chiappetta,Thomas R. Koballa (Jr.),Alfred T. Collette,1998

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