

**5.4****THE PERIODIC TABLE: ORGANIZING THE ELEMENTS**  
**SECTION REVIEW****Objectives**

- Describe the structure of the periodic table.
- Identify the position of groups, periods, and the transition metals in the periodic table.

**Key Terms**

periodic table	metalloids	post-transition metals
periods	nonmetals	halogens
periodic law	alkaline earth metals	metalloids
group	transition metals	alkaline earth metals
inert gases	lanthanides and actinides	metalloids

**Part A: Completion**

After this investigation, you will know the following properties of the elements and how to use them in your work. Each blank contains the completed sentence, so you can check your answers here.

The periodic table organizes elements into vertical \_\_\_\_\_ 1. \_\_\_\_\_ and horizontal \_\_\_\_\_ 2. \_\_\_\_\_ in order of increasing \_\_\_\_\_ 3. \_\_\_\_\_ The table is made of vertical columns called groups. Those that are similar in chemical properties are in the same \_\_\_\_\_ 4. \_\_\_\_\_ The elements in Group 1A (hydrogen) are called the \_\_\_\_\_ 5. \_\_\_\_\_ The \_\_\_\_\_ 6. \_\_\_\_\_ make up Group 0. The elements in Groups 13 and 15 are incorporated in periods 4 and 5 by the \_\_\_\_\_ 7. \_\_\_\_\_ and in the periods 6 and 7 by the \_\_\_\_\_ 8. \_\_\_\_\_ The Group 14 elements are called the \_\_\_\_\_ 9. \_\_\_\_\_ and the Group 12 elements are called the \_\_\_\_\_ 10. \_\_\_\_\_ The nonmetals of \_\_\_\_\_ 11. \_\_\_\_\_ are called the halogens. Elements with properties that are intermediate between those of metals and nonmetals are called \_\_\_\_\_ 12. \_\_\_\_\_



# Answer Key To Organizing Elements

**John T. Fodor, Gus T. Dalis**

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