

# Acids and Bases

## Overview of Acids and Bases

### Indicators: Acids and Bases

Acids: Donates its proton to base (proton donor)

Base: Donates its lone pair to acid (LBP)



Remember: (1) only H atoms, (2) only OH is a base

What about organic indicators and indicators solution?

### Acid-Base Indicators

Acids are proton donors, Bases are proton acceptors.



General acid-base reactions of acid-base reactions:

to find the strong base, the base gets protonated, change of the base

and check for weak acids/bases of the acid.

to find the weak acid, the acid gets deprotonated, change of the acid

to the other side.

Remember: -OH is a weak base, -OH<sup>-</sup> is a strong base.

1. Acid: Donates its proton to base (proton donor)

2. Base: Accepts its proton from acid (proton acceptor)

3. Conjugate acid: The acid that is left after the base has accepted its proton

4. Conjugate base: The base that is left after the acid has donated its proton

5. Strong acid: An acid that completely dissociates in water

6. Weak acid: An acid that partially dissociates in water

7. Strong base: A base that completely dissociates in water

8. Weak base: A base that partially dissociates in water

9. Amphoteric: A substance that can act as both an acid and a base

10. Buffer: A solution that resists changes in pH

11. Titration: A process used to determine the concentration of an acid or base

12. Indicator: A substance that changes color at a certain pH

13. Equilibrium: A state of balance between two or more processes

14. Le Chatelier's principle: A principle that describes how a system at equilibrium responds to changes in conditions

15. Gibbs free energy: A thermodynamic potential that can be used to predict the spontaneity of a process

16. Entropy: A measure of the disorder or randomness of a system

17. Enthalpy: A measure of the total energy of a system

18. Heat capacity: A measure of the amount of heat required to raise the temperature of a substance by a certain amount

19. Specific heat: A measure of the amount of heat required to raise the temperature of a unit mass of a substance by a certain amount

20. Latent heat: A measure of the amount of heat required to change the phase of a substance

21. Thermal conductivity: A measure of the ability of a material to conduct heat

22. Thermal expansion: A phenomenon in which the volume of a substance increases as its temperature increases

23. Thermal contraction: A phenomenon in which the volume of a substance decreases as its temperature decreases

24. Thermal stability: A measure of the ability of a substance to resist decomposition or degradation at high temperatures

25. Thermal degradation: A process in which a substance breaks down into smaller molecules as a result of heat

26. Thermal degradation products: The smaller molecules that are produced during thermal degradation

27. Thermal degradation rate: A measure of the speed at which thermal degradation occurs

28. Thermal degradation mechanism: A description of the chemical and physical processes that lead to thermal degradation

29. Thermal degradation model: A mathematical representation of the thermal degradation process

30. Thermal degradation experiment: A laboratory experiment used to study the thermal degradation of a substance

31. Thermal degradation analysis: A technique used to study the thermal degradation of a substance

32. Thermal degradation characterization: A process used to identify the thermal degradation products of a substance

33. Thermal degradation optimization: A process used to improve the thermal stability of a substance

34. Thermal degradation inhibition: A process used to slow down the thermal degradation of a substance

35. Thermal degradation acceleration: A process used to speed up the thermal degradation of a substance

36. Thermal degradation prediction: A process used to estimate the thermal degradation behavior of a substance

37. Thermal degradation monitoring: A process used to track the thermal degradation of a substance over time

38. Thermal degradation control: A process used to manage the thermal degradation of a substance

39. Thermal degradation assessment: A process used to evaluate the thermal degradation of a substance

40. Thermal degradation reporting: A process used to document the results of thermal degradation studies

41. Thermal degradation review: A process used to analyze the results of thermal degradation studies

42. Thermal degradation conclusion: A statement summarizing the findings of thermal degradation studies

43. Thermal degradation recommendation: A suggestion based on the results of thermal degradation studies

44. Thermal degradation reference: A source of information about thermal degradation

45. Thermal degradation bibliography: A list of references related to thermal degradation

46. Thermal degradation glossary: A list of terms and definitions related to thermal degradation

47. Thermal degradation index: A numerical value representing the thermal degradation of a substance

48. Thermal degradation scale: A range of values representing the thermal degradation of a substance

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# Acids And Bases Study Guide

**T. W. Graham Solomons, Craig B.  
Fryhle**



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