

- Structure, Hybridization, Resonance, Aromaticity
  - # of bonds- 1=sp (e.g. 3 bonds =  $sp^2$ )
  - Aromaticity=  $4n+2$  electrons
  - Strained ring causes greater heat of combustion
- Acids and Bases
  - Lewis acid= accepts electron pair
  - Higher percentage of s character means more acidic hydrogen
- Stereoisomerism
  - Constitutional Isomers
    - Isomers that have different atom connectivity
  - Stereoisomers
    - Same connectivity but different arrangement in space
    - Enantiomers
      - Stereoisomers that are nonsuperimposable mirror reflections
    - Diastereomers
      - Stereoisomers that are not mirror reflections of each other
  - Stereocenter=  $sp^3$  carbon bonded to four different substituents
  - Assigning R/S configuration
    - If lowest group is pointing away, determine normally
    - If lowest group is not pointing away, exchange with group pointing away and determine reverse
- Nucleophilic Substitutions and Eliminations
  - | Alkyl Halide | Strong Base | Weak Base                         | Poor Nucleophile                   |
|--------------|-------------|-----------------------------------|------------------------------------|
| 1            | $Sn2$       | $Sn2$                             | No rxn.                            |
| 2            | Mostly $E2$ | $Sn2$                             | $Sn1/E1$ in polar, protic solvents |
| 3            | $E2$        | $Sn1/E1$ in polar, protic solvent | $Sn1/E1$ in polar, protic solvents |
  - Relative strength of nucleophiles
    - Negatively charged nucleophiles react faster in substitution reactions than neutral nucleophiles
    - Nucleophilic strength is in the same order as base strength when the nucleophilic atoms are in the same period of the periodic table
    - Nucleophilic strength increases with increasing atom size when the nucleophilic atoms are in the same column of the periodic table
  - Leaving group ability
    - Increases with decreasing base strength
  - Base strength and structure

# Acs Organic Chemistry Study Guide Practice Tests

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### **Preparing for Your ACS Examination in Organic Chemistry**

Examinations Institute-American Chemical Society Division of Chemical Education,2019-12 Organic Chemistry Study Guide

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Norbert J. Pienta,Melanie M. Cooper,Thomas J. Greenbowe,2005 For courses in Methods of Teaching Chemistry Useful for new professors chemical educators or students learning to teach chemistry Intended for anyone who teaches chemistry or is learning to teach it this book examines applications of learning theories presenting actual techniques and practices that respected professors have used to implement and achieve their goals Each chapter is written by a chemist who has expertise in the area and who has experience in applying those ideas in their classrooms This book is a part of the Prentice Hall Series in Educational Innovation for Chemistry

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