

ACS Gen Chem Ch. 1 Conceptual verified question with 100% answers

1. A sodium ion differs from a sodium ion in that the sodium ion:

- a. has fewer electrons
- b. is an isotope of sodium
- c. exists only in solution
- d. has a negative charge in its nucleus

ANSWER: a. has fewer electrons

2. An atom of strontium-90 ($90/38$)Sr contains

- a. 38 electrons, 38 protons, 52 neutrons
- b. 38 electrons, 38 protons, 90 neutrons
- c. 52 electrons, 52 protons, 38 neutrons
- d. 52 electrons, 38 protons, 38 neutrons

ANSWER: a. 38 electrons, 38 protons, 52 neutrons

3. How many neutrons are in ($37/17$) Cl⁻?

- a. 17
- b. 20
- c. 21
- d. 37

ANSWER: b. 20

4. How many protons, neutrons, and electrons are present in ($47/22$)Ti³⁺

- a. 22p+, 25n⁰, 19 e⁻
- b. 22p+, 25n⁰, 25e⁻
- c. 25p+, 22n⁰, 19e⁻

Acs First Term General Chemistry Exam Answers

Oscar Krisen Buros

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Acs First Term General Chemistry Exam Answers Book Review: Unveiling the Magic of Language

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