

# Acids and Bases

## Section 19.1 Acids and Bases: An Introduction

In your textbook, read about the properties of acids and bases.

For each description below, write *acid* if it tells about a property of an acid or *base* if it tells about a property of a base. If the property does not apply to either an acid or a base, write *neither*. If it applies to both an acid and a base, write *both*.

- \_\_\_\_\_ 1. Can turn litmus paper a different color
- \_\_\_\_\_ 2. Reacts with certain metals
- \_\_\_\_\_ 3. Contains more hydrogen ions than hydroxide ions
- \_\_\_\_\_ 4. Feels slippery
- \_\_\_\_\_ 5. Reacts with carbonates
- \_\_\_\_\_ 6. Feels rough
- \_\_\_\_\_ 7. Contains equal numbers of hydrogen and hydroxide ions
- \_\_\_\_\_ 8. Tastes bitter
- \_\_\_\_\_ 9. Tastes sour

In your textbook, read about the different models of acids and bases.

Use the terms below to complete the passage. You may use each term more than once.

Arrhenius  
conjugate base

Bransted-Lowry  
hydrogen

conjugate acid  
hydroxide

The (10) \_\_\_\_\_ model of acids and bases states that an acid contains the element (11) \_\_\_\_\_ and forms ions of this element when it is dissolved in water. A base contains the (12) \_\_\_\_\_ group and dissociates to produce (13) \_\_\_\_\_ ions in aqueous solution.

According to the (14) \_\_\_\_\_ model, an acid donates (15) \_\_\_\_\_ ions, and a base accepts (16) \_\_\_\_\_ ions. According to this model, in an acid-base reaction, each acid has a (17) \_\_\_\_\_, and each base has a (18) \_\_\_\_\_.

# Acids And Bases Section 3 Answer Key

**David R. Klein**

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