

Acid-Base Equilibrium Homework Part II

Part A

1. Calculate the pH and pOH of a solution containing 7.30 grams of HCl in 1.0 L of solution.

$$7.30 \text{ g} \times \frac{1 \text{ mol}}{36.46 \text{ g}} = 0.200 \text{ mol} / 1.0 \text{ L} = 0.20 \text{ M HCl}$$

Because HCl is a strong acid, it completely dissociates, so $[\text{HCl}] = [\text{H}_3\text{O}^+] = 0.20 \text{ M}$

$$\text{pH} = -\log[0.20] = 0.70$$

$$\text{pOH} = 14 - \text{pH} = 13.30$$

2. Calculate the pH of a 0.10 M nitric acid (HNO_3) solution.

Because HNO_3 is a strong acid, the acid completely dissociates and the concentration of the acid is equal to the concentration of H^+ at equilibrium.

$$[\text{H}_3\text{O}^+] = 0.10 \text{ M}$$

$$\text{pH} = -\log(0.10) = 1$$

3. Calculate the pH of a $5.0 \times 10^{-3} \text{ M}$ solution of sodium hydroxide.

Sodium hydroxide is a strong base, which means it completely dissociates, so $[\text{NaOH}] = [\text{OH}^-]$.

$$[\text{OH}^-] = 5.0 \times 10^{-3}$$

$$\text{pOH} = -\log(5.0 \times 10^{-3}) = 1.3$$

$$\text{pH} = 14 - 1.3 = 12.7 = 13 \text{ (2 sig figs)}$$

Part B

4. Decide whether the statements are true about strong acids, weak acids, both, or neither.

- a. Always dissociates leaving behind at least some hydronium ion **both**
- b. The initial concentration of the acid is equivalent to the final concentration of hydronium ions **strong only**
- c. The pH is easily calculated **strong only**
- d. The pH must be calculated by the ICE method **weak only**
- e. The acid completely dissociates in water **strong only**
- f. The equilibrium concentration of undissociated acid is approximately 0 **strong only**
- g. K_a is very small **weak only**

5. Write the K_a expression for the following acids:

- a. $\text{HC}_2\text{H}_3\text{O}_2$



$$K_a = \frac{[\text{C}_2\text{H}_3\text{O}_2^-][\text{H}_3\text{O}^+]}{[\text{HC}_2\text{H}_3\text{O}_2]}$$

- b. H_2CO_3



$$K_a = \frac{[\text{HCO}_3^-][\text{H}_3\text{O}^+]}{[\text{H}_2\text{CO}_3]}$$

Acid Base Homework Answers

PAMELA LYNN PURO



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First Year Chemistry Students' Conceptions of Acid/base Chemistry Sally Diane Rupert, 2001 **Mapping College Chemistry** Stephen DeMeo, 2019-03-01 This text is a chemistry problem solving resource appropriate for teachers and their students who are enrolled in high school Advanced Placement Chemistry or in a first year college General Chemistry course The book incorporates a chemistry problem solving plan one that uses an innovative graphic organizer strategy The strategy successfully evaluated with students combines problem solving processes with chemical concepts that will allow students to solve the most common and difficult problems encountered in the first year of chemistry Topical problem solving will focus on limiting reactant stoichiometry identifying types of chemical reactions equilibrium acid base equilibria and electrochemistry Why would this resource be of interest to chemistry students To be successful to get into a well known college medical school physical therapy or graduate program often requires that students get an A in your pre requisite Introductory General Chemistry course To make matters worse many college professors feel that only a few students should get A grades and therefore they give difficult exams that many students fail this is the weeding out process that every pre health student is apprehensive about To succeed in this

competitive environment entails not just studying harder or longer it means re organizing textbook content so that it is meaningful to the student This is the first text of its kind to employ a reliable research based strategy that incorporates a decision based visual tool to solve chemistry textbook problems ones that can make or break a career **Organic**

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Stafford, 2012-05-16 Now updated with the latest research the breakthrough way to lose weight stop pain and prevent disease Completely updated with the latest research The Vitamin D Cure tells you all you need to know about this miraculous natural substance today s best way to heal pain prevent disease and improve your mood We now know that adding vitamin D to your daily regimen can net you unbelievable benefits from reducing your chances of having certain kinds of cancer to gaining flexibility and youthful exuberance well into your seventies and beyond And no a dose of daily sunshine doesn t give you enough Groundbreaking research reveals the healing power of this simple readily available supplement Physicians researchers and vitamin D users point to mounting evidence that the simple act of increasing the amount of vitamin D in your body can cure or help treat a remarkable number of diseases and ailments including unwanted pounds high blood pressure back pain muscle cramps obesity cancer and diabetes Leading rheumatologist and researcher Dr James Dowd reveals the causes of vitamin D deficiency and offers a simple five step program that can prevent or alleviate a host of health troubles in a matter of weeks Incorporating the latest diet and lifestyle advice the program helps you harness the power of the sunshine vitamin to transform your life Whether you struggle with arthritis or chronic pain from an old injury or you simply want to lose weight and get back to a body image that reflects your best self The Vitamin D Cure can be your key to a better more vibrant you All new recipes and meal plans Up to date information on how vitamin D helps prevent disease and reduce pain New diet and supplement breakthroughs New information from scientific studies a wealth of clinical data and case stories

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