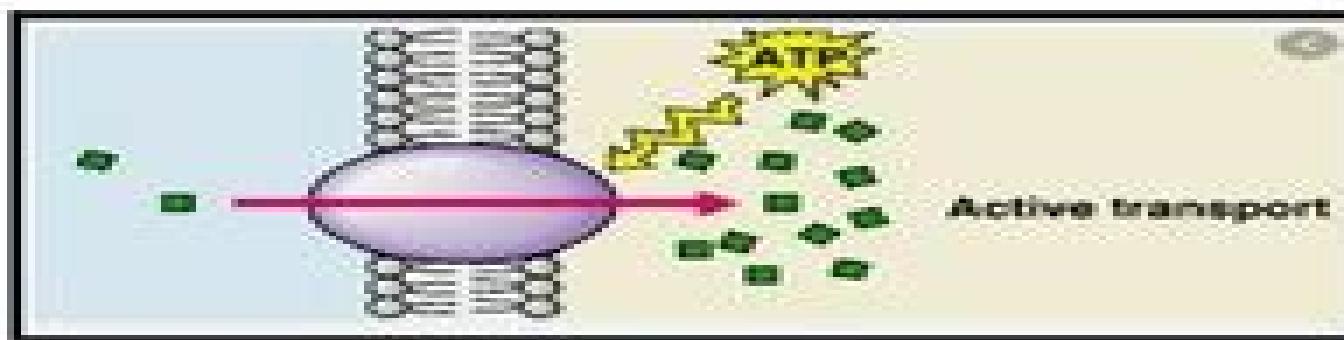




### Active Transport

Sometimes, molecules need to move into a crowded area. Since this is difficult, ATP energy is required to push the molecules into a more crowded area. See image below of active transport. Notice ATP is shown in the picture to signal that you are looking at active transport and not diffusion. Also, notice there are only 2 molecules on the left, and 10 on the right. Since this arrow is showing movement from left to right, this must be active transport for a molecule to move from low concentration to high. ATP energy is required for this to occur. Remember, ATP energy is made by catabolism.



In this image below, molecules 1 and 2 will both enter the cell because they are more crowded outside (higher concentration), so some molecules must enter to balance it out. Glucose will leave the cell because there is more glucose inside than outside.

# Active Transport Study Guide

**Kevin T. Patton, Frank B. Bell, Terry  
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Sun**

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Taylor,2013-02-01 This Student Notebook and Study Guide the ideal companion to Bruce Wingerd's The Human Body reinvents the traditional study guide by giving students a tool to help grasp information in class and reinforce learning outside of class. Too often students struggle to both learn the concepts presented and simultaneously record crucial information. The Student Notebook and Study Guide provides a structure for recording in class material that parallels the text's concept presentation and includes supplemental questions and activities for assignment outside of the classroom. A complete answer guide for both the in class and out of class materials is available online.

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December 2003 *Study Guide to Accompany Adult Health Nursing* Ruth Ann Eckenstein,Barbara L. Christensen,Elaine O. Kockrow,2002-12 Each chapter in this completely revised Study Guide promotes learning understanding and application of the core content covered in the fourth edition of the parent text Challenging exercises ensure that students meet the learning objectives set by the text Objectives are included and split into Anatomy and Physiology and Med Surg where appropriate Activities are provided and are specific to each Objective Activities may include any or all of the following Labeling and Tracing Defining Key Terms Compare and Contrast Identification NCLEX style multiple choice questions with textbook page numbers where the content is located for easy reference and a host of application type exercises Text page references are provided A complete answer key is provided in the Instructor s Manual

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web 33mol o<sub>2</sub> 6mol br<sub>2</sub> 11mol o<sub>2</sub> 18mol br<sub>2</sub> identify the limiting reactant s and excess reactant s the limiting reactant is o<sub>2</sub> since it would yield the least amount of product 18 mol br<sub>2</sub> the excess reactant is c<sub>2</sub>h<sub>3</sub>br<sub>3</sub> since its complete reaction would have yielded up to 54 mol br<sub>2</sub> think about your result

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web step 4 the reactant that produces a smaller amount of product is the limiting reagent mg mg produces less mgo mgo than does o<sub>2</sub> o<sub>2</sub> 3 98 g mgo vs 25 2 g mgo therefore mg is the limiting reagent in this reaction step 5 the reactant that produces a larger amount of product is the excess reagent

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16 16 16 16 d 4 9 16 6 e 20 36 40

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web please save your changes before editing any questions 5 minutes 1 pt viewing model 3 if 4 moles of hydrogen react with 2 moles of oxygen which of these is the excess reactant is leftover at the end and how many moles of it are leftover hydrogen 2 moles leftover hydrogen 1 mole leftover

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web if you start with 14 82 g of  $\text{Ca(OH)}_2$  and 16 35 g of  $\text{H}_2\text{SO}_4$  determine the limiting reagent b determine the number of moles of  $\text{H}_2\text{O}$  produced c determine the number of grams of  $\text{CaSO}_4$  produced d determine the number of grams of excess reagent left make sure the equation is balanced

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