

ACS ANALYTICAL CHEMISTRY STUDY GUIDE

- k_f' : $k_f' = k_f^* \cdot a$
- EDTA titrations: 1. before eq EDTA binds to metal 1:1
- 2. At eq we assume that all metal ions can bind to EDTA and then use k_f'
- Potentiometry: an analytical method in which an electric potential difference (avoltage) of a cell is measured

- Reducing agent: Donates electrons and becomes oxidized.
- oxidizing agent: Accepts electrons and becomes reduced.
- coulometry: science of measuring change and relating it to analyte
- cathode: the electrode at which reduction occurs
- Anode: the electrode at which oxidation occurs
- Ecell: $E_{cell} = E_{red} - E_{ox}$
- Faraday's Equation: $E = E - (0.0592/n) \log(a/b)$
- ||: anode always written on left of ||
- amperometry: elec techniques where the cell current is measured where pot diff between the indicator and reference electrode is controlled
- Clark electrode: Electrode for oxygen
- Biosensors: Detecting devices that can detect specific substances in cells
- amp chromo det: point, size, or charge
- limiting current: the diff between current and steady state max
- Anodic vol: Very sensative for analyzing metals into ppb
- fluroescence: used to measure singlets, atoms like rigid aromatic rings, high sensitivity
- phosphorescence: involves emission of light trip- singlet, much lower than fluro
- Transmittance: $A = -\log(T)$ or $2 - \log(T\%)$
- atomic absorbtion: lamp-flame-mono-detector
- atomic emission: flame-mono-detector
- A.A. hollow cathode lamp: used to form very narrow band lights
- A.A. atomizer: Heat sample to very high temps to break all chemical bonds, hot enough to

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(U.S.), 1980 First multi year cumulation covers six years 1965-70 Agrindex, 1993 Barron's AP Chemistry Neil D. Jespersen, Pamela Kerrigan, 2016-03-01 Extensive test preparation for the AP Chemistry exam includes Six practice AP exams three diagnostic tests and three full length practice exams that are aligned with the upcoming new AP Chemistry exam All questions answered and explained A comprehensive subject review covering the structure of matter chemical bonding states of matter physical chemistry chemical reactions and all other test topics Study tips and test taking strategies This review book can be purchased alone or with an optional CD ROM that presents two additional full length practice tests with

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