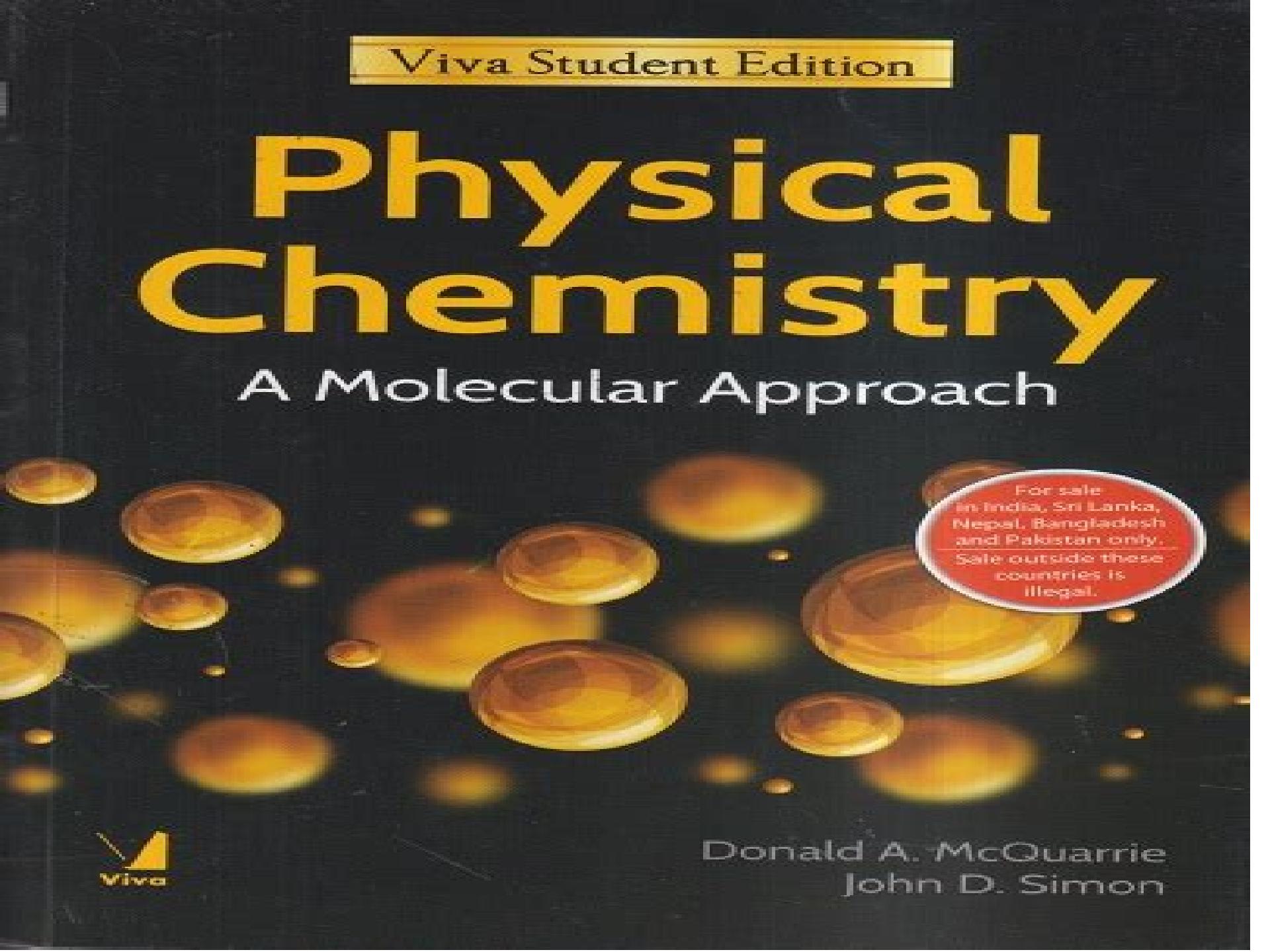


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Quanta, Matter and Change: A Molecular Approach to Physical Change Peter Atkins, Julio de Paula, Ron Friedman, 2008-11-15 Beginning with quantum mechanics introducing statistical mechanics and progressing through to thermodynamics this new text for the two semester physical chemistry course features a wealth of new applications and insights as well as new Mathematical Background inter chapters to help students review key quantitative concepts. This is a splendid book. True to the authors philosophy as outlined in the preface it approaches physical chemistry by first developing the quantum theory of molecular electronic structure then by statistical arguments moves into thermodynamics and thence to kinetics. Peter Taylor Review in Chemistry World Royal Society of Chemistry July 31 2009

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Physical Chemistry Donald A. McQuarrie, John D. Simon, 2013 *Physical Chemistry: a Molecular Approach* McQuarrie Donald A, Simon John D, 1988

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physical chemistry is taught by bridging the gap between the traditional solve a bunch of equations for a very simple model approach and the computational methods that are used to solve research problems While some recent textbooks include exercises using pre packaged Hartree Fock DFT calculations this is largely limited to giving students a proverbial black box The DIY do it yourself approach taken in this book helps student gain understanding by building their own simulations from scratch The reader of this book should come away with the ability to apply and adapt these techniques in computational chemistry to his or her own research problems and have an enhanced ability to critically evaluate other computational results This book is mainly intended to be used in conjunction with an existing physical chemistry text such as McQuarrie Simon s Physical Chemistry A Molecular Approach but it is also well suited as a stand alone text for upper level undergraduate or intro graduate computational chemistry courses Assumes no computational background Enables students to build simulations from scratch to reproduce famous literature calculations Teaches a variety of computational numerical simulation methods applicable to solving chemical problems Designed to play well with McQuarrie Simon s landmark P CHEM text but can be used with others as well

Thermodynamics Kept Simple - A Molecular Approach Roland Kjellander,2015-08-28

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Molecular Approach of Modern Physical Chemistry Ved Prakash Patial,2016 [Chemistry a Molecular Approach](#)

Nivaldo J. Tro,2011 **Basic Physical Chemistry** E Brian Smith,2012-06-26 This elegant book provides a student friendly introduction to the subject of physical chemistry It is concise and more compact than standard textbooks on the subject and it emphasises the two important concepts underpinning physical chemistry quantum mechanics and the second law of thermodynamics The principles are challenging to students because they both focus on uncertainty and probability The book explains these fundamental concepts clearly and shows how they offer the key to understanding the wide range of chemical phenomena including atomic and molecular spectra the structure and properties of solids liquids and gases chemical equilibrium and the rates of chemical reactions

Chemistry Nivaldo J. Tro,Travis David Fridgen,Lawton Shaw,2019-02-25

This innovative pedagogically driven text explains difficult concepts in a student oriented manner The book offers a rigorous and accessible treatment of general chemistry in the context of relevance Chemistry is presented visually through multi level images macroscopic molecular and symbolic representations helping students see the connections among the formulas symbolic the world around them macroscopic and the atoms and molecules that make up the world KEY TOPICS Units of Measurement for Physical and Chemical Change Atoms and Elements Molecules Compounds and Nomenclature Chemical Reactions and Stoichiometry Gases Thermochemistry The Quantum Mechanical Model of the Atom Periodic Properties of the Elements Chemical Bonding I Lewis Theory Chemical Bonding II Molecular Shapes Valence Bond Theory

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