

Stoichiometry: Multiple Choice

1. The limiting reactant of a chemical reaction is:

- A) the reactant of a chemical process that is not consumed completely
- B) the product of a chemical process that is consumed completely
- C) the product of a chemical process that is not consumed completely
- D) the reactant of a chemical process that is consumed completely
- E) none of the above

ANS: D

2. Consider the following reaction: $\text{Ca}(\text{OH})_{2(\text{aq})} + \text{Al}_2(\text{SO}_4)_{3(\text{aq})} \rightarrow \text{CaSO}_4(\text{aq}) + \text{Al}(\text{OH})_{3(\text{aq})}$

When balanced, the mole ratio for calcium hydroxide reacting with aluminum sulphate is

- A) 1:1
- B) 2:1
- C) 2:3
- D) 1:2
- E) 3:1

ANS: E

3. Sodium phosphate, Na_3PO_4 , can be prepared by combining sodium hydroxide and hydrogen phosphate as shown in the following balanced chemical equation:



If 2.5 mol of H_2PO_4 are reacted, the number of moles of water that will be produced is:

- A) 1.0
- B) 2.5
- C) 3.0
- D) 5.0
- E) 7.5

ANS: E

4. Potassium chloride and oxygen gas are formed upon heating potassium chlorate. The following balanced equation shows this decomposition process:



The number of moles of KClO_3 required to produce 5.0 mol of oxygen is

- A) 1.0
- B) 2.0
- C) 3.0
- D) 3.3
- E) 5.0

ANS: D

5. Determine the number of grams of CO_2 that can be produced from 48.0 g of the hydrocarbon C_2H_6 , according to the following equation:



- A) 37.6 g
- B) 50.2 g
- C) 87.8 g
- D) 154 g
- E) 188.4 g

6. The limiting reactant in a chemical reaction is the _____ that is _____.

- A) Reactant, in excess
- B) Product, left over
- C) Reactant, used up
- D) Product, used up

ANS: C

Multiple Choice Stoichiometry Test

**James S. Mattson, Harry B.
Mark, Hubert C. MacDonald**

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Cambridge Checkpoints VCE Chemistry Units 3 and 4 2013 Roger Slade,Maureen Slade,2012-12-14 Cambridge Checkpoints VCE are updated regularly to provide you with the most up to date exam preparation available **Roadmap to the Regents** Sasha Alcott,2003 If Students Need to Know It It's in This Book This book develops the chemistry skills of high school students It builds skills that will help them succeed in school and on the New York Regents Exams Why The Princeton Review We have more than twenty years of experience helping students master the skills needed to excel on standardized tests Each year we help more than 2 million students score higher and earn better grades We Know the New York Regents Exams Our experts at The Princeton Review have analyzed the New York Regents Exams and this book provides the most up to date thoroughly researched practice possible We break down the test into individual skills to familiarize students with the test's structure while increasing their overall skill level We Get Results We know what it takes to succeed in the classroom and on tests This book includes strategies that are proven to improve student performance We provide a breakdown of the skills based on New York standards and objectives hundreds of practice questions organized by skill two complete practice New York Regents Exams in Physical Setting Chemistry **CLEP Success** Peterson's,2007-05-04 Offers advice on improving scores on the CLEP and includes reviews for five CLEP subjects and five full length practice tests with explanatory answers **Master the CLEP** Peterson's,2012-04-30 Peterson's Master the CLEP is the essential prep guide for non traditional students who are looking to begin a college career and for current students who want to save on tuition costs by testing out of certain courses When you use this guide you will get a basic overview of the CLEP a look at the general examinations and the vast number of subject specific exams tips for registering and preparing for the CLEP and a guide to interpreting your test scores This is followed by extensive reviews of the five CLEP general examinations with pre and post tests and subject matter overviews that will help you pinpoint your strengths and weaknesses The Appendix will go into detail about a number of CLEP subject exams **Journal of Chemical Education** ,1926 Includes Report of New England Association of Chemistry Teachers and Proceedings of the Pacific Southwest Association of Chemistry Teachers **Tests** Richard C. Sweetland,Daniel J. Keyser,1983 Descriptions of over 3000 tests in English intended as a guide for psychologists educators and other personnel who need test information to meet their assessment needs Topical arrangement under 3 main sections of psychology education and business Each entry gives test name associated personal names intended age group purpose description time range scoring cost and publisher Indexes by test titles authors publishers visually impaired tests and scoring services **Optimizing STEM Education With Advanced ICTs and Simulations** Levin, Ilya,Tsybulsky, Dina,2017-06-05 The role of technology in educational settings has become increasingly prominent in recent years When utilized effectively these tools provide a higher quality of learning for students Optimizing STEM Education With Advanced ICTs and Simulations is an innovative reference source for the latest scholarly research on the integration of digital tools for enhanced

STEM based learning environments Highlighting a range of pivotal topics such as mobile games virtual labs and participatory simulations this publication is ideally designed for educators professionals academics and students seeking material on emerging educational technologies **The ETS Test Collection Catalog** Educational Testing Service. Test Collection, 1993

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Three Cognitive Skills in Chemistry and Their Application to Stoichiometry Ardas Ozsogomonyan, 1977 **A Summary of Science Education**, 1988 Leopold E.

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Table of Contents Multiple Choice Stoichiometry Test

1. Understanding the eBook Multiple Choice Stoichiometry Test
 - The Rise of Digital Reading Multiple Choice Stoichiometry Test
 - Advantages of eBooks Over Traditional Books
2. Identifying Multiple Choice Stoichiometry Test
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Multiple Choice Stoichiometry Test
 - User-Friendly Interface
4. Exploring eBook Recommendations from Multiple Choice Stoichiometry Test
 - Personalized Recommendations
 - Multiple Choice Stoichiometry Test User Reviews and Ratings
 - Multiple Choice Stoichiometry Test and Bestseller Lists
5. Accessing Multiple Choice Stoichiometry Test Free and Paid eBooks
 - Multiple Choice Stoichiometry Test Public Domain eBooks
 - Multiple Choice Stoichiometry Test eBook Subscription Services
 - Multiple Choice Stoichiometry Test Budget-Friendly Options
6. Navigating Multiple Choice Stoichiometry Test eBook Formats
 - ePUB, PDF, MOBI, and More
 - Multiple Choice Stoichiometry Test Compatibility with Devices
 - Multiple Choice Stoichiometry Test Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Multiple Choice Stoichiometry Test
 - Highlighting and Note-Taking Multiple Choice Stoichiometry Test
 - Interactive Elements Multiple Choice Stoichiometry Test

8. Staying Engaged with Multiple Choice Stoichiometry Test
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Multiple Choice Stoichiometry Test
9. Balancing eBooks and Physical Books Multiple Choice Stoichiometry Test
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Multiple Choice Stoichiometry Test
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Multiple Choice Stoichiometry Test
 - Setting Reading Goals Multiple Choice Stoichiometry Test
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Multiple Choice Stoichiometry Test
 - Fact-Checking eBook Content of Multiple Choice Stoichiometry Test
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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