

POST-LAB QUESTIONS:

1. You are given the following carbohydrates: arabinose, glucose, fructose, maltose, sucrose, and starch.
Explain the results described below.
 - a. A carbohydrate solution (from the list above) gave a positive Molisch test and negative Benedict's, Barfoed's, Bial's, and Seliwanoff's tests. When treated with hydrochloric acid and boiled for several minutes, the solution showed positive Benedict's, Barfoed's, and Seliwanoff's tests and a negative Bial's test. What carbohydrate was in the original solution?
- The carbohydrate in the original solution was **Starch**.
 - b. A solution (from the list above) containing only one carbohydrate gave a Cu_2O precipitate with Benedict's reagent. Which possible carbohydrates are present in the solution?
- The possible carbohydrates present in the solution are either **glucose or fructose or maltose or arabinose**.
 - c. To define more precisely which carbohydrate is present in the solution you analysed another sample of the same solution and it failed to give a Cu_2O precipitate with Barfoed's reagent after 5 min of boiling. Which carbohydrate is present in the solution?
- **Maltose** is present in the solution.
2. Describe how a solution of lactose would react toward these reagents.
 - a. **Benedict's Test:** Lactose is a reducing sugar that can react with Benedict's solution, which is used to detect the presence of aldehydes. Lactose contains a hemiacetal group that can open up to form a free aldehyde, which then reduces the copper (II) ions in Benedict's solution to copper(I) oxide, forming a brick-red or orange precipitate.
 - b. **Barfoed's Test:** Barfoed's reagent is used to distinguish between monosaccharides and disaccharides. A solution of lactose, being a reducing disaccharide composed of glucose and galactose, would react slowly with Barfoed's reagent to form a brick-red precipitate of copper (I) oxide.
 - c. **Seliwanoff's Test:** Seliwanoff's reagent is specific for detecting the presence of ketoses, such as fructose. Since lactose is a disaccharide with a ketose (fructose) and an aldose (galactose) component, it will not give a significant reaction with Seliwanoff's reagent. Seliwanoff's reagent contains resorcinol and hydrochloric acid, and when heated with a ketose sugar like fructose, it forms a cherry-red color within a short time. However, the reaction with lactose will be slower and less intense due to the aldose component (galactose).
3. What functional group(s) are present in the reducing carbohydrates?
Reducing carbohydrates contain one or more aldehyde (CHO) or ketone ($\text{C}=\text{O}$) functional groups. These functional groups are responsible for their reducing properties. In the case of aldoses, such as glucose and galactose, there is an aldehyde group present. In the case of ketoses, such as fructose, there is a ketone group present. These carbonyl groups have the ability to undergo oxidation-reduction reactions and can reduce other compounds, such as certain metal ions, resulting in the formation of colored precipitates or solutions in various chemical tests like Benedict's or Fehling's test.

Bettelheim Post Lab Answers

**University of California, Los Angeles.
Graduate School of Education**

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Laboratory Experiments for General, Organic and Biochemistry Frederick A. Bettelheim, Joseph Landesberg, 1991
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Table of Contents Bettelheim Post Lab Answers

1. Understanding the eBook Bettelheim Post Lab Answers
 - The Rise of Digital Reading Bettelheim Post Lab Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Bettelheim Post Lab Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Bettelheim Post Lab Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Bettelheim Post Lab Answers
 - Personalized Recommendations
 - Bettelheim Post Lab Answers User Reviews and Ratings
 - Bettelheim Post Lab Answers and Bestseller Lists

5. Accessing Bettelheim Post Lab Answers Free and Paid eBooks
 - Bettelheim Post Lab Answers Public Domain eBooks
 - Bettelheim Post Lab Answers eBook Subscription Services
 - Bettelheim Post Lab Answers Budget-Friendly Options
6. Navigating Bettelheim Post Lab Answers eBook Formats
 - ePUB, PDF, MOBI, and More
 - Bettelheim Post Lab Answers Compatibility with Devices
 - Bettelheim Post Lab Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Bettelheim Post Lab Answers
 - Highlighting and Note-Taking Bettelheim Post Lab Answers
 - Interactive Elements Bettelheim Post Lab Answers
8. Staying Engaged with Bettelheim Post Lab Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Bettelheim Post Lab Answers
9. Balancing eBooks and Physical Books Bettelheim Post Lab Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Bettelheim Post Lab Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Bettelheim Post Lab Answers
 - Setting Reading Goals Bettelheim Post Lab Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Bettelheim Post Lab Answers
 - Fact-Checking eBook Content of Bettelheim Post Lab Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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