

TEXTBOOK OF  
PHYSICAL CHEMISTRY

BY

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SECOND EDITION

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1960

# Basic Physical Chemistry Second Edition

**RS Peters**

## **Basic Physical Chemistry Second Edition:**

**TEXTBOOK OF PHYSICAL CHEMISTRY, SECOND EDITION** H. K. MOUDGIL,2014-10-21 This comprehensive textbook now in its second edition is mainly written as per the latest syllabi of physical chemistry of all the leading universities of India as well as the new syllabus recommended by the UGC This thoroughly revised and updated edition covers the principal areas of physical chemistry such as thermodynamics quantum chemistry molecular spectroscopy chemical kinetics electrochemistry and nanotechnology In a methodical and accessible style the book discusses classical irreversible and statistical thermodynamics and statistical mechanics and describes macroscopic chemical systems steady states and thermodynamics at a molecular level It elaborates the underlying principles of quantum mechanics molecular spectroscopy X ray crystallography and solid state chemistry along with their applications The book explains various instrumentation techniques such as potentiometry polarography voltammetry conductometry and coulometry It also describes kinetics rate laws and chemical processes at the electrodes In addition the text deals with chemistry of corrosion and nanomaterials This text is primarily designed for the undergraduate and postgraduate students of chemistry B Sc and M Sc for their course in physical chemistry Key Features Gives a thorough treatment to ensure a solid grasp of the material Presents a large number of figures and diagrams that help amplify key concepts Contains several worked out examples for better understanding of the subject matter Provides numerous chapter end exercises to foster conceptual understanding

*Basic Physical Chemistry for the Atmospheric Sciences* Peter V. Hobbs,2000-09-04 Revised and updated in 2000 Basic Physical Chemistry for the Atmospheric Sciences provides a clear concise grounding in the basic chemical principles required for studies of atmospheres oceans and earth and planetary systems Undergraduate and graduate students with little formal training in chemistry can work through the chapters and the numerous exercises within this book before accessing the standard texts in the atmospheric chemistry geochemistry and the environmental sciences The book covers the fundamental concepts of chemical equilibria chemical thermodynamics chemical kinetics solution chemistry acid and base chemistry oxidation reduction reactions and photochemistry In a companion volume entitled Introduction to Atmospheric Chemistry 2000 Cambridge University Press Peter Hobbs provides an introduction to atmospheric chemistry itself including its applications to air pollution acid rain the ozone hole and climate change Together these two books provide an ideal introduction to atmospheric chemistry for a variety of disciplines

**A Textbook of Physical Chemistry** Arthur Adamson,2012-12-02 A Textbook of Physical Chemistry Second Edition serves as an introductory text to physical chemistry Topics covered range from wave mechanics and chemical bonding to molecular spectroscopy and photochemistry ideal and nonideal gases the three laws of thermodynamics thermochemistry and solutions of nonelectrolytes The kinetics of gas phase reactions colloids and macromolecules and nuclear chemistry and radiochemistry are also discussed This edition is comprised of 22 chapters the first of which introduces the reader to the behavior of ideal and nonideal gases with particular

emphasis on the van der Waals equation The discussion then turns to the kinetic molecular theory of gases and the application of the Boltzmann principle to the treatment of molar polarization dipole and magnetic moments the phenomenology of light absorption and classical and statistical thermodynamics The chapters that follow focus on the traditional sequence of chemical and phase equilibria electrochemistry and chemical kinetics in gas phase and solution phase This book also considers wave mechanics and its applications molecular spectroscopy and photochemistry and the excited state and then concludes with an analysis of crystal structure colloid and polymer chemistry and radio and nuclear chemistry This reference material is intended primarily as an introductory text for students of physical chemistry *Principles of Physical Chemistry* Hans Kuhn, Horst-Dieter Försterling, David H. Waldeck, 2009-03-17 Principles of Physical Chemistry Second Edition uniquely uses simple physical models as well as rigorous treatments for understanding molecular and supramolecular systems and processes In this way the presentation assists students in developing an intuitive understanding of the subjects as well as skill in quantitative manipulations The unifying nature of physical chemistry is emphasized in the book by its organization beginning with atoms and molecules and proceeding to molecular assemblies of increasing complexity ending with the emergence of matter that carries information i.e. the origin of life a physicochemical process of unique importance The aim is to show the broad scope and coherence of physical chemistry

**Concise Physical Chemistry** Donald W. Rogers, 2011-03-01 This book is a physical chemistry textbook that presents the essentials of physical chemistry as a logical sequence from its most modest beginning to contemporary research topics Many books currently on the market focus on the problem sets with a cursory treatment of the conceptual background and theoretical material whereas this book is concerned only with the conceptual development of the subject Comprised of 19 chapters the book will address ideal gas laws real gases the thermodynamics of simple systems thermochemistry entropy and the second law the Gibbs free energy equilibrium statistical approaches to thermodynamics the phase rule chemical kinetics liquids and solids solution chemistry conductivity electrochemical cells atomic theory wave mechanics of simple systems molecular orbital theory experimental determination of molecular structure and photochemistry and the theory of chemical kinetics

**Basic Physical Chemistry** E Brian Smith, 2012-06-26 This elegant book provides a student friendly introduction to the subject of physical chemistry It is concise and more compact than standard textbooks on the subject and it emphasises the two important concepts underpinning physical chemistry quantum mechanics and the second law of thermodynamics The principles are challenging to students because they both focus on uncertainty and probability The book explains these fundamental concepts clearly and shows how they offer the key to understanding the wide range of chemical phenomena including atomic and molecular spectra the structure and properties of solids liquids and gases chemical equilibrium and the rates of chemical reactions *Physical Chemistry* William M. Davis, 2011-12-06 Designed specifically for a two semester introductory course sequence in physical chemistry this text presents core principles and topics Straightforward and

streamlined it presents the necessary amount of detail for comprehension Organized in such a way that the various topics covered are connected to each other it allows students to see physical chemistry as an interconnected discipline and not a series of unrelated concepts Each chapter in this new edition has been thoroughly updated and includes new information on computational applications more end of chapter problems and new chapters on nanotechnology and surface chemistry

***The Journal of Physical Chemistry***,1903    **Fundamental Principles of Physical Chemistry ... Revised (second Edition. (Fifth Printing.))**. Carl Frederic PRUTTON (and MARON (Samuel Herbert)),1956    **A Textbook Of Physical Chemistry (Vol. 2) (3 Edition)** K L Kapoor,2000-02 This is one of the four volumes that provides a thorough understanding of the principles and basic concepts of physical chemistry Emphasis is placed on applications of the principles This volume deals with thermo dynamics and chemical equilibrium    **Physical Chemistry. (Second Edition, Revised.)**. Arthur James Mee,1939    *The Journal of Physical Chemistry* Wilder Dwight Bancroft,1922 Includes section New Books    **The Pharmaceutical Era** ,1918    **Science** ,1925 Vols for 1911 13 contain the Proceedings of the Helminothological Society of Washington ISSN 0018 0120 1st 15th meeting    **London, Edinburgh and Dublin Philosophical Magazine and Journal of Science** ,1921    **Philosophical Magazine** ,1919    *Merck's Report* Theodore Weicker,1926    **The Quarterly Review of Biology** Raymond Pearl,1928 Includes section New biological books and other bibliographies    **The School World** ,1900    *Chemical News and Journal of Physical Science* William Crookes,James H. Gardiner,Gerald Druce,H. W. Blood-Ryan,1913

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