

CHEMISTRY UNIT 1 STUDY GUIDE.

What is Matter?

- Solid, Liquid, gas
- Anything that takes up space & has mass
- Mass is the amount of matter in an object.

Types

- Substance- a particular kind of matter — pure - made up of elements and compounds
- Mixture- more than one kind of matter. 2+ elements and compounds

Properties:

Physical – property that can be observed and measured without changing the substance.

Chemical – property that can only be observed by changing the type of substance.

Extensive- only depends on the amount of matter

Mass, weight, volume, length

Intensive – only depends on the type of matter, not the amount used to identify a substance.

Color, Odor, melting/ freezing point.

Ductility- the ability to make copper into wires.

States of Matter:

	Definite volume	Definite shape	Temperature increase	compressible
SOLID	Yes	Yes	Small expans	No
LIQUID	Yes	No Takes up shape of container	Small expans	No
GAS	No	No	Large expans	Yes



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Lingsheng Yao

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Directing Study of High School Pupils Maxie Nave Woodring,Mrs. Cecile Woodard (White) Flemming,Mrs. White Fleming,1929 **Directory of Distance Learning Opportunities** Modoc Press, Inc.,2003-02-28 This book provides an overview of current K 12 courses and programs offered in the United States as correspondence study or via such electronic delivery systems as satellite cable or the Internet The Directory includes over 6 000 courses offered by 154 institutions or distance learning consortium members Following an introduction that describes existing practices and delivery methods the Directory offers three indexes Subject Index of Courses Offered by Level Course Level Index Geographic Index All information was supplied by the institutions Entries include current contact information a description of the institution and the courses offered grade level and admission information tuition and fee information enrollment periods delivery information equipment requirements credit and grading information library services and accreditation

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