

# Practice

## 8.1 Worksheet

# Practice 8 1 Worksheet

**Marion W. Stuart,Lwuk715**



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web record the atomic mass and abundance in nature of certain isotopes explain how abundance in nature influences

average atomic mass note this activity requires access to the phet interactive check it out here isotopes atomic mass

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web nov 13 2016 are all atoms of an element the same how can you tell one isotope from another use the sim to learn about isotopes and how abundance relates to the average atomic mass of an element

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web benefit that sim in students about isotopes and instructions abundance related to one average atomic mass of into element are all atoms of certain element the same how can you tell one isotope from another

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web use the link phet colorado en simulation isotopes and atomic mass to go to the phet website for the simulation isotopes and atomic mass begin with the isotopes and atomic mass click on the symbols next to the bar with symbol and abundance in nature to display the data

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web multiple choice 1 minute 1 pt lithium has only two stable isotopes use the sim to determine the average atomic mass of a sample containing three lithium 6 atoms and two lithium 7 atoms 6 51556 amu 6 61565 amu 6 34875 amu 6 41548 amu

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web write at least three conclusions from your exploration 1 protons and neutrons determine the mass of the atom 2 only protons determine the atomic number 3 the more unstable the atom the less presence or no presence in nature 1 protons and neutrons determine the mass of the atom 2 only protons determine the atomic number 3

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web microsoft word phet isotopes and atomic mass author lauralt2 created date 9 28 2017 9 36 15 am

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web isotopes and atomic mass model 1 make isotopes open the isotopes and atomic mass simulation play with the make isotopes tab of the simulation for a few minutes and then answer the following questions 1 what particles determine the mass number number of protons and neutrons 2 why is mass number always a whole

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web this guided inquiry activity has two learning goals students will be able to students will be able to 1 explain the difference between atomic mass and mass number and 2 calculate average atomic mass from percent abundance and isotopic mass

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web find the average atomic mass of an element given the abundance and mass of its isotopes predict how the mass and name of an isotope will change given a change in the number of protons neutrons or electrons

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web oct 22 2020 this a screencast video tutorial on how to answer and navigate the phet simulator named isotopes and atomic mass found at phet colorado edu sims ht