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(C.B.S.E. Syllabus)

Class XI



AVICHAL PUBLISHING COMPANY

Apc Chemistry Lab Manual Class 11

**Neena Sinha, R Rangarajan, R P
Manchanda, R K Gupta, Rajesh Kumar**

Apc Chemistry Lab Manual Class 11:

Practical/Laboratory Manual Chemistry Class - XI Er. Meera Goyal, 2021-05-29

1 Basic Laboratory Techniques

1 To cut a glass tube or glass rod
2 To bend the glass rod at an angle
3 To draw a glass jet from a glass tube
4 To bore a cork and fit a glass tube into it
Viva Voce

2 Characterisation and Purification of Chemical Substances

1 To determine the melting point of the given unknown organic compound and its identification simple laboratory technique
Viva Voce
2 To determine the boiling point of a given liquid when available in small quantity simple laboratory method
Viva Voce
3 To prepare crystals of pure potash alum $K_2SO_4 \cdot Al_2(SO_4)_3 \cdot 24H_2O$ from the given impure sample
4 To prepare the pure crystals of copper sulphate from the given crude sample
5 To prepare pure crystals of benzoic acid from a given impure sample
Viva Voce

3 Measurement of pH Values

1 To determine the pH value of vegetable juices fruit juices tap water and washing soda by using universal pH paper
2 To determine and compare the pH values of solutions of strong acid HCl and weak acid CH_3COOH of same concentration
3 To study the pH change in the titration of strong base Vs strong acid by using universal indicator paper
4 To study the pH change by common ion CH_3COO^- ion in case of weak acid CH_3COOH
5 To determine the change in pH value of weak base NH_4OH in presence of a common ion NH_4^+
Viva Voce

4 Chemical Equilibrium

1 To study the shift in equilibrium between ferric ions and thiocyanate ions by changing the concentrations of either of the ions
2 To study the shift in equilibrium between $Co(H_2O)_6^{2+}$ and Cl^- ions by changing the concentrations of either of the ions
Viva Voce

5 Quantitative Analysis

1 To prepare M 10 oxalic acid solution by direct weighing method
2 To prepare M 10 solution of sodium carbonate by direct weighing method
3 To determine the strength of given solution of sodium hydroxide by titrating it against N 10 or M 20 solution of oxalic acid
4 To determine the strength of a given solution of hydrochloric acid by titrating it against a standard N 10 or M 20 sodium carbonate solution
Viva Voce

6 Qualitative Analysis

Analysis of Anions
Analysis of Cations
Viva Voce

7 Detection of Elements in Organic Compounds

1 To detect the presence of nitrogen sulphur and halogens in a given organic compound by Lassaigne's test
2 To detect the presence of nitrogen sulphur and halogens in the given organic compound sample number by Lassaigne's test
Viva Voce

INVESTIGATORY PROJECTS

1 Checking of Bacterial Contamination in Water

1 To check the bacterial contamination in drinking water by testing sulphide ions
Viva Voce

2 Methods of Water Purification

1 To purify water from suspended impurities by using sedimentation
2 To purify water by boiling
3 To purify water by distillation method
4 To purify water by reverse osmosis technique
5 To purify water by GAC method
6 To purify water by bleach treatment
7 To purify water by oxidising agent
8 To purify water by ozone treatment method
Viva Voce

3 Water Analysis

1 To test the hardness of different water samples
Viva Voce

4 Foaming Capacity of Various Soaps

1 To compare the foaming capacity of different washing soaps
2 To study the effect of addition of sodium carbonate on foaming capacity of washing soap
Viva Voce

5 Tea Analysis

1 To study the acidity of different samples of tea leaves tea by using pH paper
Viva Voce

6 Analysis of Fruits and Vegetable Juices

1 To analysis the fruit and vegetable juices for the constituent present in them

Viva Voce 7 Rate of Evaporation 1 To study the rate of evaporation of different liquids IViva Voce 8 Effect of Acids and Bases on Tensile Strength of Fibres 1 To compare the tensile strength of natural fibres and synthetic fibres 2 To study the effect of acids and bases on tensile strength of different fibres Viva Voce

Chemistry Lab Manual Class XI | Follows the Latest CBSE Syllabus and Other State Board Following the CBSE Curriculum. Edugorilla Prep Experts,2022-07-05 Theories are incomplete without experimentation So in order to enhance the practical skills of students the CBSE conducts the practical exams for class 11 chemistry along with theoretical exams to test the scientific aptitude of students CBSE Class 11th chemistry Lab Manual has been prepared strictly according to CBSE guidelines which follows the NCERT and other state board curriculum All the experiments in this Lab Manual have been given in a simple lucid and interactive manner The theory given with each experiment gives complete knowledge of each term concept and definition in order to provide a compact knowledge to the students at one place Viva Voce questions are given with each experiment to test students understanding of related experiments Mr Kaushalesh Dwivedi is the author of this complete lab activity book for class 11 chemistry All the lab based experiments have well labeled diagrams which clearly demonstrate the details of the apparatus that are used the correct way of handling Laboratory apparatus and perform the experiments methodically The basic purpose of this Lab Manual is to provide the students with appropriate guidance so that they can easily understand carry out and document the experiments in the Science Laboratory The Lab Manual comes with practical lessons followed by lab experiments after each chapter projects and Notebook for jotting down all the essential information this develops a better understanding about the concept behind each class 11 chemistry theory there is an experiment which brings authentic and easy to retain conclusions through its observations and which is verified on a real time basis The aim of this Science Lab Manual is to help class 11 students to prepare and understand the complex chemistry concepts through experimental approach by utilizing pedagogically enhanced content of the Lab Manual and with a compact knowledge base for improving their analytical experimental skills and learning abilities

Chemistry Lab Manual Neena Sinha, R Rangarajan, R P Manchanda, R K Gupta, Rajesh Kumar, Lab Manual **General Chemistry I Laboratory Manual, 11E - CHM 111** Brightpoint Community College,2023-05-26 *Current Index to Journals in Education* ,1977 Acp Chem 1172 General Chemistry Lab Manual Brooks/Cole,2015-11-11 **The Software Encyclopedia** ,1986 Acp Chem 1171 General Chemistry Lab Manual Brooks/Cole,2014-04-25 *Popular Photography* ,1990-11 **Nutrition Abstracts and Reviews** ,1998 *General College Chemistry I and II (CHM 11/12) Laboratory Manual* Darrell D. Ebbing,2017 **Literature on Information Retrieval and Machine Translation** Service Bureau Corporation,1959 *Chemistry AP Lab Manual 8E* Steven S. Zumdahl,2009-03-19 **CHEM 111 Lab Manual General Chemistry** Christofield,2018-08-13 *Chemistry* Peter T. Wassell,2009-08-10 **Government reports annual index** ,199? GENERAL CHEMISTRY LAB MANUAL FOR SCIENCE MAJORS. EDWARD. WITTEN,2022 **Who's Who in the South and Southwest, 1986-1987** ,1986-08

Chem 111-112 Larry Alan Scheich, Tammy Janine Melton, New Mexico State University. Department of Chemistry and Biochemistry, 2002 *Who's who in Finance and Industry*, 1998

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university acceptance rate is 70 75 university accepts local and international students language of education is english
students can apply for bachelor b sc masters m sc and for phd degree programmes

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