

ACS Biochemistry Final

pKa (chapter 3)

measure of the tendency of a group to give up a proton (acidity); tendency decreases tenfold as pKa increases by one unit

thioester bond (1)

compounds with functional group C-S-CoA-C (eg, in acetyl-CoA)

zwitterion (3)

dipolar ion with both positive and negative groups but overall neutral charge; can act as either an acid or base

isoelectric point (pi) (chapter 3)

the characteristic pH at which the net electric charge is zero

SDS - Sodium Dodecyl Sulfate (chap 3)

detergent used to unfold proteins and give them uniform negative charge

SDS Page (3)

chromatography used to separate proteins based on mass. light proteins travel fast than heavier ones

PCR - Polymerase chain reaction (3)

copies DNA multiple times to increase sample size

isoelectric focusing (chap 3)

procedure used to determine the isoelectric pt (pi) of a protein. Protein migrates through gel until pH = pi (net charge = 0)

Two-Dimensional Electrophoresis (3)

combines isoelectric focusing and SDS electrophoresis; separates proteins by both molecular weight and pi

specific activity (3)

number of enzyme units per mg of total protein (a measure of enzyme purity)

activity (3)

total units of a certain enzyme in a solution

Peptide bonds (chapter 4)

C-N bond with double bond character due to resonance (C-N bond cannot rotate, and is planar)

Edman degradation (3)

used in the sequencing of polypeptides; labels and removes ONLY the amino-residue from a polypeptide. carried out in a machine called a sequenator

ϕ in peptide bonding (chapter 4)

angle around the α -carbon - amide nitrogen bond

ψ in peptide bonding (chapter 4)

angle around the α -carbon - carbonyl carbon bond

Ramachandran Plot (4)

shows favorable ϕ - ψ angle combinations. 3 main "wells" for α -helices, β -sheets, and left handed α -helices

Levinthal's Paradox (4)

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**Kristen Murphy,Olga Michels,Kyle
Willian**

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1. Understanding the eBook Acs Exam Study Guide Biochemistry
 - The Rise of Digital Reading Acs Exam Study Guide Biochemistry
 - Advantages of eBooks Over Traditional Books
2. Identifying Acs Exam Study Guide Biochemistry
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Acs Exam Study Guide Biochemistry
 - User-Friendly Interface
4. Exploring eBook Recommendations from Acs Exam Study Guide Biochemistry
 - Personalized Recommendations
 - Acs Exam Study Guide Biochemistry User Reviews and Ratings
 - Acs Exam Study Guide Biochemistry and Bestseller Lists
5. Accessing Acs Exam Study Guide Biochemistry Free and Paid eBooks
 - Acs Exam Study Guide Biochemistry Public Domain eBooks
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 - Acs Exam Study Guide Biochemistry Budget-Friendly Options

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 - Acs Exam Study Guide Biochemistry Compatibility with Devices
 - Acs Exam Study Guide Biochemistry Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Acs Exam Study Guide Biochemistry
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 - Interactive Elements Acs Exam Study Guide Biochemistry
8. Staying Engaged with Acs Exam Study Guide Biochemistry
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Acs Exam Study Guide Biochemistry
9. Balancing eBooks and Physical Books Acs Exam Study Guide Biochemistry
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Acs Exam Study Guide Biochemistry
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Acs Exam Study Guide Biochemistry
 - Setting Reading Goals Acs Exam Study Guide Biochemistry
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Acs Exam Study Guide Biochemistry
 - Fact-Checking eBook Content of Acs Exam Study Guide Biochemistry
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

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