

- Structure, Hybridization, Resonance, Aromaticity
 - # of bonds- 1=sp (e.g. 3 bonds = sp^2)
 - Aromaticity= $4n+2$ electrons
 - Strained ring causes greater heat of combustion
- Acids and Bases
 - Lewis acid= accepts electron pair
 - Higher percentage of s character means more acidic hydrogen
- Stereoisomerism
 - Constitutional Isomers
 - Isomers that have different atom connectivity
 - Stereoisomers
 - Same connectivity but different arrangement in space
 - Enantiomers
 - Stereoisomers that are nonsuperimposable mirror reflections
 - Diastereomers
 - Stereoisomers that are not mirror reflections of each other
 - Stereocenter= sp^3 carbon bonded to four different substituents
 - Assigning R/S configuration
 - If lowest group is pointing away, determine normally
 - If lowest group is not pointing away, exchange with group pointing away and determine reverse
- Nucleophilic Substitutions and Eliminations
 - | Alkyl Halide | Strong Base | Weak Base | Poor Nucleophile |
|--------------|-------------|-----------------------------------|------------------------------------|
| 1 | $Sn2$ | $Sn2$ | No rxn. |
| 2 | Mostly $E2$ | $Sn2$ | $Sn1/E1$ in polar, protic solvents |
| 3 | $E2$ | $Sn1/E1$ in polar, protic solvent | $Sn1/E1$ in polar, protic solvents |
 - Relative strength of nucleophiles
 - Negatively charged nucleophiles react faster in substitution reactions than neutral nucleophiles
 - Nucleophilic strength is in the same order as base strength when the nucleophilic atoms are in the same period of the periodic table
 - Nucleophilic strength increases with increasing atom size when the nucleophilic atoms are in the same column of the periodic table
 - Leaving group ability
 - Increases with decreasing base strength
 - Base strength and structure

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Creating this article collection will foster collaboration between chemical education researchers and teachers of organic chemistry. This opportunity allows these instructors to share evidence based practices, experiences, challenges and innovative approaches from CER literature and beyond. This Research Topic bridges discipline based education research and the scholarship of teaching and learning which will help advance organic chemistry education and improve student outcomes.

ACS Organic Chemistry Exams - The Official Guide Examinations Institute, 2002 *Analytical methods, formation mechanisms and control strategies for endogenous hazardous substances produced during the thermal processing of foods*
Hongshun Yang, Jie Zheng, Hao Dong, 2023-03-01 [Test List](#) Cornell University. Testing and Service Bureau, 1950

Student Study Guide and Solutions Manual, Organic Chemistry, Fifth Edition William Henry Brown, Brent L. Iverson, Christopher S. Foote, Eric Anslyn, 2008 The best way for students to learn organic chemistry concepts is to work relevant and interesting problems on a daily basis. Authored by Brent and Sheila Iverson, The University of Texas at Austin, this comprehensive manual offers detailed solutions to all in text and end of chapter problems. It helps students achieve a deeper intuitive understanding of the material through constant reinforcement and practice, ultimately resulting in much better preparation for in class quizzes and tests as well as national standardized tests such as the DAT and MCAT.

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