

Acid Base test review Sheet

1. List three properties of acids: _____
 2. List three properties of bases: _____
 3. Acids turn litmus _____ and bases turn it _____
 4. Name the following acids: Put a check next to it if it is a strong acid
 HBr _____ H_2SO_4 _____ HClO_4 _____
 HNO_3 _____ HI _____ H_2S _____
 HNO_2 _____
 6. What is the formula for carbonic acid? _____ chloric acid _____ hydrochloric acid _____
 8. What is the formula for finding the pH of an acid? _____
 9. What is the $[\text{H}_3\text{O}^+]$ of a neutral solution? _____ what is the $[\text{OH}^-]$ of a neutral solution? _____
 10. What is the K_w of water? _____
- Define :
11. Arrhenius Acid _____
 12. Arrhenius base _____
 13. Bronsted-Lowry acid _____
 14. Bronsted-Lowry base _____
 15. Lewis acid _____
 16. Lewis base _____
 17. indicator _____
 18. titration _____

19. Fill in the following table:

$[\text{H}^+]$	pH	$[\text{OH}^-]$	pOH	Acid or base
10^{-11}				
	5.6			
		2.27×10^{-5}		
			8.33	

20. The strongest acid would have a pH of _____, the strongest base would have a pH of _____
21. What are the products of a neutralization reaction? _____
22. Write the balanced equation for the neutralization of $\text{Ca}(\text{OH})_2$ and HCl _____

Acid And Base Test Review Sheet Answers

Darrell D. Ebbing



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