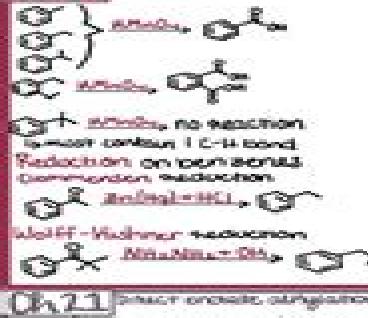


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Ch 17 Oxidation of aldehydes

$\text{C}_6\text{H}_5\text{CHO} \xrightarrow[\text{H}_2\text{O}]{\text{CuO}, \text{H}_2\text{O}_2} \text{C}_6\text{H}_5\text{COOH} + \text{Cu}^{2+}$

$\text{C}_6\text{H}_5\text{CHO} \xrightarrow{\text{AgNO}_2, \text{H}_2\text{O}_2, \text{CH}_3\text{COOH}} \text{C}_6\text{H}_5\text{COOH} + \text{Ag}_2\text{O}$

Reduction using LiAlH₄

$\text{C}_6\text{H}_5\text{CHO} \xrightarrow[\text{H}_2\text{O}]{\text{LiAlH}_4} \text{C}_6\text{H}_5\text{CH}_3$

acid $\xrightarrow{\text{LiAlH}_4(\text{excess}), \text{H}_2\text{O}}$ ester

chloride

Using DIBAL-H

$\text{C}_6\text{H}_5\text{CHO} \xrightarrow[\text{H}_2\text{O}]{\text{DIBAL-H}} \text{C}_6\text{H}_5\text{CH}_3$ ester

Hydrogenation of Aldehydes • Ketones

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