

# Approach to Arterial Blood Gases (ABGs)

1. Adequate oxygenation?

Normal ABG on Room Air = 7.40 / 40 / 90 / 24

Normal  $\text{pCO}_2$  should be  $\text{pCO}_2 \approx 4-5$  ( $\text{pCO}_2$  in room air = 30)

$\text{pCO}_2 < 5$ , then expected

Possible hypoxemia

2. Define the acid-base disturbance

$\text{pH} < 7.40 =$  Acidosis

$\text{pH} > 7.40 =$  Alkalosis

See slide on  
Hypoxemia, Pathogenesis and Clinical Findings

3. Identify the primary ( $\Delta^+$ ) process

4. Compensation mechanisms in lungs and kidneys attempt to keep/retain  $\text{CO}_2$  or  $\text{HCO}_3^-$  to maintain normal pH. Determination of compensation is appropriate (appropriate normal ratio of change in  $\text{pCO}_2$  :  $\text{HCO}_3^-$  is shown in green boxes). If inappropriate, there is a secondary ( $\Delta^-$ ) process accounting

$\Delta^+ \text{ HCO}_3^-$

$\Delta^+ \text{ HCO}_3^-$

$\Delta^- \text{ HCO}_3^-$

5. Calculate anion gap (AG) to determine the presence and type of metabolic acidosis. This is done regardless of the  $\Delta^+$  or  $\Delta^-$  process, as there may also be a hidden metabolic acidosis

Calculate  $\text{AG} = \text{Na}^+ - \text{Cl}^- - \text{HCO}_3^-$  (normal is  $\approx 12$ )

6. HAGMA: refer to normal blood buffer system, acid gain should match bicarbonate loss. If not, identify if another process is causing an inappropriate loss or gain of  $\text{HCO}_3^-$ .  
 $\text{Gain of acid (AG)} = \text{AG} - 12$   
 $\text{Loss of } \text{HCO}_3^- (\text{AHCO}_3^-) = 24 - \text{HCO}_3^-$

Classically, the compensating anion is  $\text{Cl}^-$ , maintaining a normal AG

If a nonbicarbonate anion is present,  $\Delta^- \text{ HCO}_3^-$  is compensated by  $\Delta^+$  other serum anions to maintain neutral charge

Where the cause is addition of an abnormal acid,  $\text{HCO}_3^-$  is used up as a buffer, and the compensating anion is the abnormal acid's conjugate base - which is not measured in the AG calculation

$\Delta^- \text{ HCO}_3^-$  is compensated by unmeasured anions ( $\Delta^- \text{ the AG} - \text{f}$ )

7. Calculate Cation Gap (net positive) or Urine net charge (for HAGMA) to narrow the etiology

$\text{AG} = 12$

$\text{AG} > 12$

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\* 10th or 11th year of publication

Legend:

Diagnose (Red)

Mechanism

Lab Finding/Calculation

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See slide on Normal Anion Gap Metabolic Acidosis: Pathogenesis and Laboratory Findings

See slide on High Anion Gap Metabolic Acidosis: Pathogenesis and Laboratory Findings

# Abmg Study Guide

**Diane L. Baker, Jane L. Schuette, Wendy  
R. Uhlmann**

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## **Table of Contents Abmg Study Guide**

1. Understanding the eBook Abmg Study Guide

- The Rise of Digital Reading Abmg Study Guide
- Advantages of eBooks Over Traditional Books

2. Identifying Abmg Study Guide

- Exploring Different Genres
- Considering Fiction vs. Non-Fiction
- Determining Your Reading Goals

3. Choosing the Right eBook Platform

- Popular eBook Platforms
- Features to Look for in an Abmg Study Guide
- User-Friendly Interface

4. Exploring eBook Recommendations from Abmg Study Guide

- Personalized Recommendations
- Abmg Study Guide User Reviews and Ratings
- Abmg Study Guide and Bestseller Lists

5. Accessing Abmg Study Guide Free and Paid eBooks

- Abmg Study Guide Public Domain eBooks
- Abmg Study Guide eBook Subscription Services
- Abmg Study Guide Budget-Friendly Options

6. Navigating Abmg Study Guide eBook Formats

- ePub, PDF, MOBI, and More
- Abmg Study Guide Compatibility with Devices
- Abmg Study Guide Enhanced eBook Features

7. Enhancing Your Reading Experience

- Adjustable Fonts and Text Sizes of Abmg Study Guide
- Highlighting and Note-Taking Abmg Study Guide
- Interactive Elements Abmg Study Guide

8. Staying Engaged with Abmg Study Guide

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Abmg Study Guide

9. Balancing eBooks and Physical Books Abmg Study Guide
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Abmg Study Guide
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Abmg Study Guide
  - Setting Reading Goals Abmg Study Guide
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Abmg Study Guide
  - Fact-Checking eBook Content of Abmg Study Guide
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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