

Section 5.1 Organizing the Elements

(pages 126–129)

This section explains how Mendeleev organized elements into a periodic table. It also discusses the predictions he made about undiscovered elements and how the discovery of those elements supported the use of the table of the elements.

Reading Strategy (page 126)

Identifying Main Ideas: As you read, complete the table by identifying the main idea for each topic. For more information on this reading strategy, see the **Reading and Study Skills** in the **Skills and Reference Handbook** at the end of your textbook.

Topic	Main Idea
Mendeleev's proposal	Mendeleev arranged the elements into rows in order of increasing mass so that elements with similar properties were in the same columns.
Mendeleev's prediction	Mendeleev used the position of existing elements to predict properties of nonexistent elements.
Evidence supporting Mendeleev's table	The close match between Mendeleev's predictions and the actual properties of new elements showed that Mendeleev's table was correct.

The Search for Order (page 126)

- Is the following sentence true or false? The first elements to be identified were mainly gases. False
- As the number of known elements grew, so did the need to organize them into groups based on their properties.
- Circle the letter of each category that the French chemist Antoine Lavoisier used to classify elements.

a. gases
 b. metals
 c. liquids

a. metals
 b. nonmetals

Mendeleev's Periodic Table (pages 127–129)

- Is the following sentence true or false? Mendeleev needed to organize information about 63 elements. True
- Mendeleev's strategy for classifying elements was modeled on none of these.
- Circle the letter of each type of information Mendeleev knew about each element.

a. name
 b. number of protons
 c. relative mass
 d. properties

Answer Key Study Guide Organizing The Elements

**Patricia A. Potter,Anne G. Perry,Kyla
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