

When the surface of the Earth, this acceleration due to gravity is 9.8 m/s^2 and is directed downward. In English units, what force due to gravity acts on a person who weighs 150 pounds?

When a car is moving at a constant velocity, the net force acting on it is zero.

$$F = ma$$

When a car is moving at a constant velocity, the net force acting on it is zero.

Students will work with several models of the following: A person (or) remains at rest when pushed. When pushed, it will move but will come to rest when friction acts on it. When riding in a car, the motion of the car is velocity as the velocity. In the driver's seat, the driver will continue to move forward until your seat belt stops you and stops you.

Week 14, Day 4

When a car is moving at a constant velocity, the net force acting on it is zero. A constant force will move forward the way of the object and the magnitude of the force will increase over time.



In this diagram, the net force is equal to the force of the car pushing on the car without the frictional force. The net force is equal to the force of the car pushing on the car without the frictional force. The net force is equal to the force of the car pushing on the car without the frictional force.

1. If a molecule has 1 sulfur (S) atom and 2 fluorine (F) atoms, what is its chemical formula?
2. What is formed in the chemical reaction $2H_2 + O_2 \rightarrow 2H_2O$? Which side of the equation has the reactants, the left side or the right side? Which side of the equation has the products, the left side or the right side?

3. What is the most stable electron configuration for atoms?
 - a. Every energy level has an equal number of electrons.
 - b. The highest energy level has a full set of electrons.
 - c. There is an even number of electrons.
4. Which two of the following ways can an atom achieve a stable electron configuration?
 - a. Sharing electrons with another atom.
 - b. Transferring electrons to another atom.
 - c. Adding energy to its existing atoms.
5. Which type of bond forms when two nonmetal atoms share valence electrons?
 - a. ionic bond
 - b. covalent bond
 - c. hydrogen bond

6. When two atoms have an unequal pull on the electrons they share, the bond is (polar/nonpolar) covalent bond.
7. Which of the following bond types are the weakest?
 - a. ionic bond
 - b. covalent bond
 - c. hydrogen bond
8. Which phase of water is denser, solid or liquid?

9. Which property of water is not a result of hydrogen bonding?
 - a. surface tension
 - b. transparency
 - c. cohesion
 - d. adhesion

10. A water spider insect is able to walk on water, despite being denser than water, because water has hydrogen bonds. This results in a phenomenon called (capillary action/surface tension), so the insect will not sink.

ADVANCED CONCEPTS

11. In the following chemical equation, how many molecules of water are formed as a product?

$$2H_2 + O_2 \rightarrow 2H_2O$$
12. In the chemical equation in question 11, how many molecules of oxygen are present as a reactant?
13. Verify how many atoms of each element are present on each side of the following equation, and determine which element or elements are not balanced.

$$NO_2 \rightarrow 2NO + O_2$$
14. How would you balance the equation in question 13?

Answer Guide For Compass Learning Physical Science

Samuel Austin Allibone



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