

Name: ANSWER KEY

## Chemistry Assessment

1) Explain why a compound is considered a pure substance.

A compound is considered a pure substance because pure substances contain only one substance, with no others mixed in. Pure substances contain only one type of particle and can be either elements or compounds. A pure substance made of 2 or more elements bonded together is a compound, which can be broken down into elements again.

2) Identify if the following examples are physical or chemical changes:

| Example  | Physical or chemical change | Evidence  |
|--|-----------------------------|---|
| Alka Seltzer tablet is dropped into water and begins to fizz | This is a chemical change   | With the evidence of fizz forming, this indicates that gas bubbles are being created. This would be difficult to reverse.                     |
| Juice crystals are stirred into water                        | This is a physical change   | With the evidence of the juice crystals dissolving into smaller crystals, you can still see the juice crystals once the water has evaporated. |
| A glow stick starts glowing                                  | This is a chemical change   | Chemicals in glow not react together to create light. The release of light energy is evidence of chemical change.                             |
| Ice cube is dropped into a glass of soda pop                 | This is a physical change   | When you see ice melting from a solid to liquid in soda, this is a change of state, which is a physical change.                               |

# Assessment Of Basic Chemistry Concepts Answer Sheet

**Albert A Gayle**

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