

## Experiment 5: Charles' Law

### Data Sheet



	Trial 1	Trial 2	Trial 3
Temperature of hot water ( $^{\circ}\text{C}$ )	78 $^{\circ}\text{C}$	80 $^{\circ}\text{C}$	84 $^{\circ}\text{C}$
Temperature of hot water (K)	351.15	353.15	357.15
Temperature of ice + water ( $^{\circ}\text{C}$ )	16 $^{\circ}\text{C}$	14 $^{\circ}\text{C}$	10 $^{\circ}\text{C}$
Temperature of ice + water (K)	289.15	287.15	283.15
Volume of flask (mL)	92.6	93.2	96.0
Volume of water entering the flask (mL)	13.2	12.8	14.6

9

### Calculation

1. Calculate the volume of cold air theoretically.
2. Calculate the percentage of error in your experiment.

# Answer Key For Charles Law Data Sheet

**John Indermaur, Charles Thwaites**



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