

## Compounds

Word	Definition
Binary compound	A compound that consists of two elements.
Dipole attractions	The attraction of the $\delta+$ end of one polar molecule to the $\delta-$ end of another polar molecule.
Dipole moment	An arrow along the line of symmetry in a polar molecule that shows the net direction that electrons are being pulled towards the partially negative end of the molecule.
Electronegativity difference	The difference in electronegativity between two elements in a bond.
Electrical conductivity (metals)	The ability of a substance to allow electrons to pass from atom to atom through itself from a source of electricity to an electrical ground.
Electrolyte	A solution containing dissolved ions that can conduct electricity through it.
Empirical formula	The simplest whole-number mole ratio of elements in a compound, used to write the formulas of ionic compounds.
Formula mass	The sum of the atomic masses of an element or compound, measured in grams/mole. Reported to the nearest tenth of a gram per mole.
Hydrogen bonds	The strong attraction of the $\delta+$ end of one polar molecule to the $\delta-$ end of another polar molecule. The two ends form temporary covalent bonds.
Intermolecular Attractive Force	The forces that hold molecules together in the solid and liquid phase. These are the forces that must be overcome to melt or boil a substance. Also called "van der Waal's forces".
Ionic compound	Compounds consisting of a metal and a nonmetal that are ionically bonded in a whole-number ratio.
London dispersion force	The weak attractive force caused by temporary dipoles in nonpolar molecules.
Metallic bond	A bond formed between metal atoms as they collectively share their conducting electrons evenly between metal kernels.
Molecular formula	The actual number of nonmetal atoms in a molecule, a whole-number multiple of the empirical formula.
Molecule	A particle made of nonmetal atoms covalently bonded together to form a distinct particle.
Network solid	A crystal lattice formed from covalently bonded nonmetal atoms with no distinct molecules.
Nonpolar molecule	A molecule with symmetrical electron distribution resulting in no polar bonds cancelling each other out to yield no partially charged ends.
Percent composition	The formula mass of an element divided by the formula mass of the compound that contains the element multiplied by 100.
Polar molecule	A molecule with asymmetrical electron distribution resulting in partially charged ends.
Polyatomic ion	An ion formed by atoms bonding together in such a way that a net charge is formed.
Ternary compound	A compound that consists of three or more elements, usually containing a polyatomic ion.

# Answer Key 2011 Mark Rosengarten

**Maria Romeralo, Sandra  
Baldauf, Ricardo Escalante**

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*Dictyostelids* Maria Romeralo, Sandra Baldauf, Ricardo Escalante, 2013-07-16 Since their discovery in 1869 the dictyostelids have attracted the attention of scientists in a wide variety of fields. This interest has stemmed from their peculiar lifestyle and developmental properties which were shaped by the evolutionary forces that generated multicellularity during eukaryotic evolution. More recently the dictyostelids have gained attention due to the striking similarities found at the genomic, cellular and biochemical levels with human cells which has propelled the species *Dictyostelium discoideum* to become a model system for biology and medicine in many laboratories. This book covers the latest advances in our knowledge of these extraordinary organisms with topics spanning from their evolutionary history, ecology and diversity to the recent discoveries regarding their cellular and molecular biology.

[The Paralysis of Analysis in African American Studies](#) Stephen Ferguson II, 2023-09-21 Stephen C. Ferguson II provides a philosophical examination of Black popular culture for the first time. From extensive discussion of the philosophy and political economy of Hip Hop music through to a developed exploration of the influence of the postmodernism poststructuralist ideology on African American studies, he argues how postmodernism ideology plays a seminal role in justifying the relationship between corporate capitalism and Black popular culture. Chapters cover topics such as cultural populism, capitalism and Black liberation, the philosophy of Hip Hop music and Harold Cruse's influence on the cultural turn in African American studies. Ferguson combines case studies of past and contemporary Black cultural and intellectual productions with a Marxist ideological critique to provide a cutting edge reflection on the economic structure in which Black popular culture emerged. He highlights the contradictions that are central to the juxtaposition of Black cultural artists as political participants in socioeconomic struggle and the political participants who perform the rigorous task of social criticism. Adopting capitalism as an explanatory framework, Ferguson investigates the relationship between postmodernism as social theory, current manifestations of Black popular culture and the theoretical work of Black thinkers and scholars to demonstrate how African American studies have been shaped.

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