

Section 1: Study Guide

As you work through Section 1 of the Introduction to Chemistry Laboratory Experiments, let the questions help guide your understanding and identify what you know. Answer the questions to your best ability as we build understanding over time.

Please note that these questions are not meant to replace laboratory safety or other procedures.

Will have questions? Check out our [FAQ section](#).

Section 1: Chemical Structures and Periodic Trends

Lesson 1: Identification of Matter and Matter Changes

Learning objective: The student identifies the composition of matter as is physical and chemical change.

Class goals:

1. Define and provide an example for matter [why here](#) listed for Lesson 1, on the Introduction List?
2. Identify the class of matter given a substance's relative chemical formula(s) (ionic substance, element, compound, homogeneous mixture, heterogeneous mixture)?
3. Identify the state of matter (solid, liquid, gas) given a description of physical properties, include one or more of the following?
 - a. Amount of kinetic energy (How hot the particles are moving)
 - b. Strength of the intermolecular forces between particles
 - c. Organization (ordered vs. disordered)
 - d. Compressible (or not) (or ability to flow) ability to be compressed
4. Identify the state of matter given the melting point and boiling point (freezing point) of a substance and the standard temperature?
5. Identify and classify changes/processes of matter as either chemical or physical?
6. Identify the formula for all physical and chemical changes?
7. Identify a property and related substance or substances?

Lesson 2: Measurements in Chemistry

Answers Study Guide Chemistry Section 1

Linda D. Williams



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