

Part – 1

Chapter 1 – Basic Concepts

- 1) To find the number of molecules in a given compound which of the following is employed?
 a) Chromatography
 c) Kc
 (NET-2015)

b) Le-Chatelier's principle
 d) None of these
- 2) What is the percentage of N in NH_4NO_3 ?
 a) 40%
 (NET-2015)

b) 50%
 c) 35%
 d) 30%
- 3) Mass spectrometer can be used to find:
 a) Mass of atom
 c) Charge to mass ratio of atom
 (NET-2013)

b) Mass of ion
 d) Charge to mass ratio of ion
- 4) 4 gram atoms of carbon has mass of:
 a) 12
 (NET-2013)

b) 24
 c) 48
 d) 56
- 5) 0.5 mole of SO_2 has how much mass of oxygen?
 a) 32
 (NET-2013)

b) 12
 c) 16
 d) None of these
- 6) In combustion analysis method CO_2 is absorbed by?
 a) $\text{Mg}(\text{OH})_2$
 (NET-2013)

b) KOH
 c) $\text{Mg}(\text{ClO}_3)_2$
 d) None of these
- 7) 1 amu is?
 a) $1.66 \times 10^{-27} \text{ kg}$
 (NET-2013)

b) $1.602 \times 10^{-19} \text{ kg}$
 c) $1.66 \times 10^{-27} \text{ kg}$
 d) None of these
- 8) Mass of neutron in amu is?
 a) 1.008386
 (NET-2013)

b) 1.007376
 c) 0.0005
 d) None of these
- 9) Mass spectrometer is used to determine which of the following quantity?
 a) Atomic masses
 (NET-2013)

b) Atomic number
 c) Both a and b
 d) None of these
- 10) Chemical properties of an atom depends upon number of
 a) Electrons
 (NET-2013)

b) Neutrons
 c) Protons
 d) Nucleons
- 11) One mole of Oxygen has _____ number of atoms as compared to one mole of Nitrogen.
 a) Equal
 (NET-2013)

b) Half
 c) Double
 d) Quarter

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