

5.4**THE PERIODIC TABLE: ORGANIZING THE ELEMENTS**
SECTION REVIEW**Objectives**

- Describe the structure of the periodic table.
- Identify the position of groups, periods, and the transition metals in the periodic table.

Key Terms

periodic table	metalloids	post-transition metals
periods	nonmetals	halogens
periodic law	alkaline earth metals	metalloids
group	transition metals	alkaline earth metals
inert gases	lanthanides and actinides	metalloids

Part A: Completion

After their completion, complete the blank spaces after reading each concept and answer the questions that are located in this overview. Each blank can be completed with one, two, or three lines.

The periodic table organizes elements into vertical _____ 1. _____ and horizontal _____ 2. _____ in order of increasing _____ 3. _____ The table is made of vertical columns or groups. Each column is similar in chemical properties and is often named _____ 4. _____ The elements in Group 1A (hydrogen) are called the _____ 5. _____ The _____ 6. _____ make up Group 0. The elements in Groups 13 and 15 are incorporated in periods 4 and 5 by the _____ 7. _____ and in the periods 6 and 7 by the _____ 8. _____ The Group 14 elements are called the _____ 9. _____ and the Group 12 elements are called the _____ 10. _____ The nonmetals of _____ 11. _____ are called the halogens. Elements with properties that are intermediate between metals and nonmetals are called _____ 12. _____



Answer Key To Organizing Elements

Charles S. Wasson

Answer Key To Organizing Elements:

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