

Name: ANSWER KEY

## Chemistry Assessment

An example of a compound pure substance is  $\text{CO}_2$  or Distilled water as pure  $\text{H}_2\text{O}$ .

- 1) Explain why a compound is considered a pure substance.

A compound is considered a pure substance because pure substances contain only one substance, with no others mixed in. Pure substances contain only one type of particle and can be either elements or compounds. A pure substance made of 2 or more elements bonded together is a compound, which can be broken down into elements again.

- 2) Identify if the following examples are physical or chemical changes:

Example	Physical or chemical change	Evidence
Alka Seltzer tablet is dropped into water and begins to fizz	This is a chemical change	with the evidence of fizz forming, this indicates that gas bubbles are being created. This would be difficult to reverse.
Juice crystals are stirred into water	This is a physical change	with the evidence of the juice crystals dissolving into smaller crystals, you can still see the juice crystals once the water has evaporated.
A glow stick starts glowing	This is a chemical change.	chemicals in glow rods react together to create light. The release of light energy is evidence of chemical change.
Ice cube is dropped into a glass of soda pop	This is a physical change	when you see ice melting from a solid to liquid in soda, this is a change of state, which is a physical change.

# Assessment Of Basic Chemistry Concepts Answer Sheet

**D Keegan**



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