

# SYNTHESIS OF ASPIRIN

## I. OBJECTIVES AND BACKGROUND

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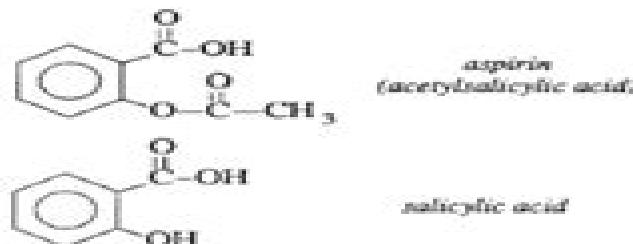
- synthesize acetylsalicylic acid (aspirin) by carrying out a simple organic reaction.
- separate your product from the reaction mixture by vacuum filtration.
- purify your product by recrystallization.
- perform a chemical test to identify the change in functional group from reactant to product, and
- determine the success of your synthesis by calculating the percentage yield of your product.

## INTRODUCTION

Aspirin is one of the most widely used medications in the world. It is employed as an analgesic (pain relief), an anti-pyretic (fever control) and an anti-inflammatory. More recently, studies have indicated that daily intake of small doses of aspirin can lower the risk of heart attack and stroke in high-risk patients.

The history of aspirin and its precursor dates back to ancient times. Documents attributed to Hippocrates, the father of modern medicine, from the 4th century B.C. refer to the alleviation of pain by chewing on the bark of a willow tree or ingesting a powder made from the bark and leaves of the willow. This remedy was passed on from generation to generation.

Fast forward now to the 19th century, where the field of organic chemistry began to experience tremendous growth. By 1838, chemists had managed to isolate, purify and identify the component of willow bark that provided the analgesic benefit. The compound was named salicylic acid, which was based on the genus name of the willow.



# Aspirin Synthesis Lab Manuals

**William L. Masterton, Stanley M.  
Cherim**

## Aspirin Synthesis Lab Manuals:

**Laboratory Manual for Principles of General Chemistry** Jo Allan Beran,2010-11-01 This new edition of the Beran lab manual emphasizes chemical principles as well as techniques The manual helps students understand the timing and situations for the various techniques The Beran lab manual has long been a market leading lab manual for general chemistry Each experiment is presented with concise objectives a comprehensive list of techniques and detailed lab intros and step by step procedures

**Laboratory Manual of Medicinal Chemistry III** Dr. Niranjan S. Mahajan,Dr. Shrinivas D. Joshi,Dr. T. Prabha,Mr. Shrikrishna S. Maske, We are very pleased to put forth the Laboratory Manual of Medicinal Chemistry III This manual is prepared as per PCI B Pharm course regulations 2014 and is divided into three sections for laboratory techniques determination of oil values and preparations of organic compounds The methods of all the experiments are added from the recent research papers so that the advancement in the methods or apparatus can be addressed This manual is designed for outcome based education and each experiment is arranged in a uniform way such as practical significance practical outcomes PrOs and its mapping with course outcomes theory resources used procedure precautions observations result conclusion references and synopsis questions Each experiment offers an opportunity to perform practical work allowing students to gain proficiency in effectively managing equipment handling glassware chemicals and reagents and writing analytical reports In addition questions are provided at the end of the experiments to enhance students knowledge which will be beneficial for them as they pursue higher studies During the laboratory period you will have to multitask while you are doing experiment It is essential to document properly what you do and what you observe while doing the practical Always plan your work ahead and think about what you are doing why you are doing it what is happening and what you can conclude from your experiment We acknowledge the help and co operation extended by various persons in bringing out this manual We are highly indebted to the authors of various books and articles mentioned in the reference which became a major source of information for writing this manual We also thank the publishers designers and printers who graciously worked hard to publish this manual in time We hope that this manual will assist students in understanding concepts principles and performing procedures We wish you all the best

*Laboratory Manual for Principles of General Chemistry* J. A. Beran,Mark Lassiter,2022-08-16 Laboratory Manual for Principles of General Chemistry 11th Edition covers two semesters of a general chemistry laboratory program The material focuses on the lab experiences that reinforce the concepts that not all experimental conclusions are the same and depend on identifying an appropriate experimental procedure selecting the proper apparatus employing the proper techniques systematically analyzing and interpreting the data and minimizing inherent variables As a result of good data a scientific and analytical conclusion is made which may or may not be right but is certainly consistent with the data Experiments write textbooks textbooks don t write experiments A student s scientific literacy grows when experiences and observations associated with the scientific method are encountered Further

experimentation provides additional cause effect observations leading to an even better understanding of the experiment The 11th edition s experiments are informative and challenging while offering a solid foundation for technique safety and experimental procedure The reporting and analysis of the data and the pre and post lab questions focus on the intuitiveness of the experiment The experiments may accompany any general chemistry textbook and are compiled at the beginning of each curricular unit An Additional Notes column is included in each experiment s Report Sheet to provide a space for recording observations and data during the experiment Continued emphasis on handling data is supported by the Data Analysis section

**Laboratory Manual** Jo A. Beran,1990 **Laboratory Manual to Accompany Chemistry in Context**

American Chemical Society,2005-02 The 5th edition Laboratory Manual that accompanies Chemistry in Context is compiled and edited by Gail Steehler Roanoke College The experiments use microscale equipment wellplates and Beral type pipets as well as common materials Project type and cooperative collaborative laboratory experiments are included Additional experiments are available on the Online Learning Center as is the instructor s guide

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Navidi Arthur D. Baker,1990 **Laboratory Manual** Harold Hart,1988 **Introduction to Chemistry, Laboratory**

**Manual** T. R. Dickson,1994-12-23 Teaches chemistry by offering a dynamic provocative and relevant view of the topic and its importance to society and our daily lives Three themes are stressed throughout the text developing chemical thinking and a chemical vision learning problem solving methods and utilizing group work and discussion activities These themes involve and engage the students in their own learning processes they are challenged to be active The presentation of topics has been altered to include a new chapter which introduces the students to scientific thinking and shows that chemistry involves interesting and relevant topics The reorganization presents many core concepts in the first five chapters preparing students for later chapters In addition the author has added vignettes throughout the chapters referring to health technology the environment and society as well as to specific tools of direct use to students

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Timberlake,1979 **Laboratory Manual of Elementary Organic Chemistry** George Holmes Richter,1951 **The**

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**The Extraordinary Chemistry of Ordinary Things, Laboratory Manual** Carl H. Snyder,1997-09-18 Shows how chemistry affects our lives To emphasize the experimental basis of chemistry chapters begin with demonstrations that readers can perform for themselves Think Speculate Reflect and Ponder sections include questions that ask readers to think

critically about the connections between chemistry society and individual values      [Instructors Manual to Lab Manual](#) Ralph Petrucci, William Harwood, Geoffrey Herring, 2001      **Laboratory Experiments for General Chemistry** Harold R. Hunt, Toby F. Block, George M. McKelvy, 1998      **Laboratory Manual for Fundamentals of General, Organic, and Biological Chemistry** John R. Holum, Ruth C. Denison, 1978

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web slovenia s l ou ' v i: n i ə s l ə sloh vee nee ə slovene slovenija slo'vè:nija officially the republic of slovenia slovene republika slovenija abbr rs is a country in southern central europe it is bordered by italy to the west austria to the north hungary to the northeast croatia to the south and southeast and a short coastline within the