

## BrainPOP- pH Scale

Name \_\_\_\_\_

Visit the main page of the **Kid Zone** at <http://sciencespot.net/> to find the link for BrainPOP.

1. The term \_\_\_\_\_ refers to the chemical potential of hydrogen.
2. The pH \_\_\_\_\_ measures how \_\_\_\_\_ or basic a substance on a scale of 1 to 14. \_\_\_\_\_ measure from 1 to 7, while the \_\_\_\_\_ or alkaline side measures from 7 to 14.
3. Identify each as an acid (A) or a base (B).

<input type="checkbox"/> Lemon juice	<input type="checkbox"/> Soap	<input type="checkbox"/> Battery acid	<input type="checkbox"/> Hydrochloric acid
<input type="checkbox"/> Toothpaste	<input type="checkbox"/> Vinegar	<input type="checkbox"/> Baking soda	<input type="checkbox"/> Floor cleaner
4. Really strong acids and bases are found at the \_\_\_\_\_ of the scale, while those near the \_\_\_\_\_ are weaker. Right in the middle at exactly 7, or pH \_\_\_\_\_, you have pure water.
5. Acids contain \_\_\_\_\_ ions ( $H^+$ ), while bases contain \_\_\_\_\_ ( $OH^-$ ) ions. Ions are atoms that have either lost or gained \_\_\_\_\_. In bases you find hydroxyl ions, which have \_\_\_\_\_ electrons. Hydrogen ions found in acids are \_\_\_\_\_ that have had an electron knocked off, which is why these ions are always trying to get the lost electrons back. When you put a metal in acid, the metal starts to \_\_\_\_\_.
6. Bases have the power to \_\_\_\_\_ acids. The hydroxyl ion \_\_\_\_\_ its electron with the hydrogen ion. The  $H^+$  from the hydrogen ion and the  $OH^-$  from the hydroxyl ion bond together to form  $H_2O$ , which is \_\_\_\_\_ with pH neutral. Different types of \_\_\_\_\_ are formed depending on the types of acids and bases involved in the reaction.
7. When Moby poured the baking soda (a base) into the vinegar (an acid), a reaction occurred. Salt, water, and \_\_\_\_\_ were produced by the reaction.
8. \_\_\_\_\_ are substances that can block changes in their pH for period of time. You can find them on buffered aspirin, which keeps your \_\_\_\_\_ from dissolving the medicine too quickly. Buffers are made by combining \_\_\_\_\_ acids and bases. Buffers also are found in our body to keep our the pH of our \_\_\_\_\_ at or around 7.4.
9. Most science labs have pH paper, which is paper that has been soaked in a special chemical \_\_\_\_\_. They turn \_\_\_\_\_ in an acid and \_\_\_\_\_ in a base. By comparing the color of the pH paper to a chart, you can determine how \_\_\_\_\_ an acid or base is.
10. Constant reactions between acids and bases keep our \_\_\_\_\_ from being too acidic or too alkaline.

Word Bank:
Acidic
Acids
Atoms
Bases
Blood
Blue
Buffers
Carbon dioxide
Center
Dissolve
Electrons
Ends
Environment
Gained
Hydrogen
Hydroxyl
Indicator
Neutral
Neutralize
pH
Red
Salts
Scale
Shares
Stomach acid
Strong
Water
Weak

# Brainpop Ph Scale Answer Key

**NARAYAN CHANDER**

## Brainpop Ph Scale Answer Key:

The pH Scale Mary Griffin,2018-12-15 The pH scale measures how acidic or basic a substance is ranging from 0 to 14 Readers will learn how certain substances rank on the pH scale what happens when acids and bases are mixed and how water can make a substance either acidic or basic These significant science concepts are discussed in approachable text and supported by motivating fact boxes charts images and photographs

## ACID-BASE CHEMISTRY NARAYAN

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