

written as  $dv/dt = g$ , with solution  $v(t) = gt + v_0$ . The object is released with an initial velocity  $v_0$ .

(b). Suppose that the object is released from a height of  $h$  units above the ground. Using the fact that  $v = dx/dt$ , in which  $x$  is the *downward displacement* of the object, we obtain the differential equation for the displacement as  $dx/dt = gt + v_0$ . With the origin placed at the point of release, direct integration results in  $x(t) = gt^2/2 + v_0 t$ . Based on the chosen coordinate system, the object reaches the ground when  $x(t) = h$ . Let  $t = T$  be the time that it takes the object to reach the ground. Then  $gT^2/2 + v_0 T = h$ . Using the quadratic formula to solve for  $T$ ,

$$T = \frac{-v_0 \pm \sqrt{v_0^2 + 2gh}}{g}.$$

The *positive* answer corresponds to the time it takes for the object to fall to the ground. The *negative* answer represents a previous instant at which the object could have been launched upward (*with the same impact speed*), only to ultimately fall downward with speed  $v_0$ , from a height of  $h$  units above the ground.

(c). The impact speed is calculated by substituting  $t = T$  into  $v(t)$  in part (a). That is,  $v(T) = \sqrt{v_0^2 + 2gh}$ .

10(a,b). The general solution of the differential equation is  $Q(t) = ce^{-rt}$ . Given that  $Q(0) = 100$  mg, the value of the constant is given by  $c = 100$ . Hence the amount of thorium-234 present at any time is given by  $Q(t) = 100e^{-rt}$ . Furthermore, based on the hypothesis, setting  $t = 1$  results in  $82.04 = 100e^{-r}$ . Solving for the rate constant, we find that  $r = -\ln(82.04/100) = .19796/\text{week}$  or  $r = .02828/\text{day}$ .

(c). Let  $T$  be the time that it takes the isotope to decay to *one-half* of its original amount.

From part (a), it follows that  $50 = 100e^{-rT}$ , in which  $r = .19796/\text{week}$ . Taking the natural logarithm of both sides, we find that  $T = 3.5014$  weeks or  $T = 24.51$  days.

11. The general solution of the differential equation  $dQ/dt = -rQ$  is  $Q(t) = Q_0e^{-rt}$ , in which  $Q_0 = Q(0)$  is the initial amount of the substance. Let  $\tau$  be the time that it takes the substance to decay to *one-half* of its original amount,  $Q_0$ . Setting  $t = \tau$  in the solution, we have  $0.5Q_0 = Q_0e^{-r\tau}$ . Taking the natural logarithm of both sides, it follows that  $-r\tau = \ln(0.5)$  or  $r\tau = \ln 2$ .

# Boyce Manual

**L Darling-Hammond**



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