



LABORATORY Manual

CHEMISTRY

CBSE Class XI

Experiments
Projects
Viva-Voce



Abd Lab Manual Class 11 Chemistry

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Public Instruction**



Abd Lab Manual Class 11 Chemistry:

Practical/Laboratory Manual Chemistry Class XI based on NCERT guidelines by Dr. S. C. Rastogi & Er. Meera Goyal Dr. S. C. Rastogi , Er. Meera Goyal, 2020-06-23 An Excellent Book in Accordance with the latest syllabus for Class 11 Prescribed by CBSE NCERT and Adopted by Various State Education Boards

A Basic Laboratory Techniques

- 1 To cut a glass tube or glass rod
- 2 To bend the glass rod at an angle
- 3 To draw a glass jet from a glass tube
- 4 To bore a cork and fit a glass tube into it

B Characterisation and Purification of Chemical Substances

- 1 To determine the melting point of the given unknown organic compound and its identification simple laboratory technique
- 2 To determine the boiling point of a given liquid when available in small quantity simple laboratory method
- 3 To prepare crystals of pure potash alum $K_2SO_4 \cdot Al_2(SO_4)_3 \cdot 24H_2O$ from the given impure sample
- 4 To prepare the pure crystals of copper sulphate from the given crude sample
- 5 To prepare pure crystals of benzoic acid from a given impure sample

C Measurement of pH Values

- 1 To determine the pH value of vegetable juices fruit juices tap water and washing soda by using universal pH paper
- 2 To determine and compare the pH values of solutions of strong acid HCl and weak acid CH_3COOH of same concentration
- 3 To study the pH change in the titration of strong base Vs strong acid by using universal indicator paper
- 4 To study the pH change by common ion CH_3COO^- ion in case of weak acid CH_3COOH
- 5 To determine the change in pH value of weak base NH_4OH in presence of a common ion NH_4^+

D Chemical Equilibrium

- 1 To study the shift in equilibrium between ferric ions and thiocyanate ions by changing the concentrations of either of the ions
- 2 To study the shift in equilibrium between Co^{2+} and Cl^- ions by changing the concentrations of either of the ions

E Quantitative Analysis

- 1 To prepare M 10 oxalic acid solution by direct weighing method
- 2 To prepare M 10 solution of sodium carbonate by direct weighing method
- 3 To determine the strength of given solution of sodium hydroxide by titrating it against N 10 or M 20 solution of oxalic acid
- 4 To determine the strength of a given solution of hydrochloric acid by titrating it against a standard N 10 or M 20 sodium carbonate solution

F Qualitative Analysis

- 1 Analysis of Anions
- 2 Analysis of Cations

G Detection of Elements in Organic Compounds

- 1 To detect the presence of nitrogen sulphur and halogens in a given organic compound by Lassaigne's test
- 2 To detect the presence of nitrogen sulphur and halogens in the given organic compound sample number by Lassaigne's test

INVESTIGATORY PROJECTS

A Checking of Bacterial Contamination in Water

- 1 To check the bacterial contamination in drinking water by testing sulphide ions

B Methods of Water Purification

- 1 To purify water from suspended impurities by using sedimentation
- 2 To purify water by boiling
- 3 To purify water by distillation method
- 4 To purify water by reverse osmosis technique
- 5 To purify water by GAC method
- 6 To purify water by bleach treatment
- 7 To purify water by oxidising agent
- 8 To purify water by ozone treatment method

C Water Analysis

- 1 To test the hardness of different water samples

D Foaming Capacity of Various Soaps

- 1 To compare the foaming capacity of different washing soaps
- 2 To study the effect of addition of sodium carbonate on foaming capacity of washing soap

E Tea Analysis

- 1 To study the acidity of different samples of tea leaves tea by using pH paper

F

Analysis of Fruits and Vegetable Juices 1 To analyse the fruit and vegetable juices for the constituent present in them G Rate of Evaporation 1 To study the rate of evaporation of different liquids H Effect of Acids and Bases on Tensile Strength of Fibres 1 To compare the tensile strength of natural fibres and synthetic fibres 2 To study the effect of acids and bases on tensile strength of different fibres Log Antilog Table

Chemistry Lab Manual Class XI | Follows the Latest CBSE Syllabus and Other State Board Following the CBSE Curriculum. Edugorilla Prep Experts, 2022-07-05

Theories are incomplete without experimentation So in order to enhance the practical skills of students the CBSE conducts the practical exams for class 11 chemistry along with theoretical exams to test the scientific aptitude of students CBSE Class 11th chemistry Lab Manual has been prepared strictly according to CBSE guidelines which follows the NCERT and other state board curriculum All the experiments in this Lab Manual have been given in a simple lucid and interactive manner The theory given with each experiment gives complete knowledge of each term concept and definition in order to provide a compact knowledge to the students at one place Viva Voce questions are given with each experiment to test students understanding of related experiments Mr Kaushalesh Dwivedi is the author of this complete lab activity book for class 11 chemistry All the lab based experiments have well labeled diagrams which clearly demonstrate the details of the apparatus that are used the correct way of handling Laboratory apparatus and perform the experiments methodically The basic purpose of this Lab Manual is to provide the students with appropriate guidance so that they can easily understand carry out and document the experiments in the Science Laboratory The Lab Manual comes with practical lessons followed by lab experiments after each chapter projects and Notebook for jotting down all the essential information this develops a better understanding about the concept behind each class 11 chemistry theory there is an experiment which brings authentic and easy to retain conclusions through its observations and which is verified on a real time basis The aim of this Science Lab Manual is to help class 11 students to prepare and understand the complex chemistry concepts through experimental approach by utilizing pedagogically enhanced content of the Lab Manual and with a compact knowledge base for improving their analytical experimental skills and learning abilities

The Cumulative Book Index ,1913 **The United States Catalog** Mary Burnham, Carol Hurd, 1928 *Journal of Industrial and Engineering Chemistry* ,1921 **The Journal of Industrial and Engineering Chemistry** ,1921 The United States Catalog ,1913 Chemical Abstracts ,1927 *Oregon School Directory* Oregon. Office of Superintendent of Public Instruction, 1927 **Research in Education** ,1968-07 **Who's who in America** ,1924 **The University of Virginia Record** University of Virginia, 1916 **Resources in Education** ,1984 **Otto E. Miller, Plaintiff-Respondent, Against Fred W. Smythe, Defendant-Appellant** , The Publishers Weekly ,1911 **Calendar** University of Saskatchewan, 1946 *Catalogue ...* West Virginia University, 1919 **Catalog** James Millikin University, 1922 **Chemical News and Journal of Physical Science** ,1903 **Annual Circular of the Illinois Industrial University** University of Illinois (Urbana-Champaign campus), 1906

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Table of Contents Abd Lab Manual Class 11 Chemistry

1. Understanding the eBook Abd Lab Manual Class 11 Chemistry
 - The Rise of Digital Reading Abd Lab Manual Class 11 Chemistry
 - Advantages of eBooks Over Traditional Books
2. Identifying Abd Lab Manual Class 11 Chemistry
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Abd Lab Manual Class 11 Chemistry
 - User-Friendly Interface
4. Exploring eBook Recommendations from Abd Lab Manual Class 11 Chemistry
 - Personalized Recommendations
 - Abd Lab Manual Class 11 Chemistry User Reviews and Ratings
 - Abd Lab Manual Class 11 Chemistry and Bestseller Lists
5. Accessing Abd Lab Manual Class 11 Chemistry Free and Paid eBooks
 - Abd Lab Manual Class 11 Chemistry Public Domain eBooks
 - Abd Lab Manual Class 11 Chemistry eBook Subscription Services
 - Abd Lab Manual Class 11 Chemistry Budget-Friendly Options
6. Navigating Abd Lab Manual Class 11 Chemistry eBook Formats

- ePub, PDF, MOBI, and More
 - Abd Lab Manual Class 11 Chemistry Compatibility with Devices
 - Abd Lab Manual Class 11 Chemistry Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Abd Lab Manual Class 11 Chemistry
 - Highlighting and Note-Taking Abd Lab Manual Class 11 Chemistry
 - Interactive Elements Abd Lab Manual Class 11 Chemistry
 8. Staying Engaged with Abd Lab Manual Class 11 Chemistry
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Abd Lab Manual Class 11 Chemistry
 9. Balancing eBooks and Physical Books Abd Lab Manual Class 11 Chemistry
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Abd Lab Manual Class 11 Chemistry
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Abd Lab Manual Class 11 Chemistry
 - Setting Reading Goals Abd Lab Manual Class 11 Chemistry
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Abd Lab Manual Class 11 Chemistry
 - Fact-Checking eBook Content of Abd Lab Manual Class 11 Chemistry
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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