

## CHEM 101 ACS Study Guide

## Chapter 1

**SQ-1.** Planck's constant is  $6.626 \times 10^{-34}$  J·s. What is this value in J·μs?

(A)  $6.626 \times 10^{-17}$  J·μs      (B)  $6.626 \times 10^{-19}$  J·μs  
 (C)  $6.626 \times 10^{-14}$  J·μs      (D)  $6.626 \times 10^{-16}$  J·μs

**SQ-4.** Twenty irregular pieces of an unknown metal have a collective mass of 28.225 g. When carefully placed in a graduated cylinder that originally contained 7.75 mL of water, the final volume read 11.70 mL. What is the density of the metal (in g·cm<sup>-3</sup>)?

(A) 2.41 g·cm<sup>-3</sup>      (B) 3.64 g·cm<sup>-3</sup>      (C) 7.15 g·cm<sup>-3</sup>      (D) 11.2 g·cm<sup>-3</sup>

**PQ-12.** If a palladium nanoparticle has a density of 12.0 g·cm<sup>-3</sup>, what is the mass of a nanoparticle with a volume of  $1.54 \times 10^{-21}$  cm<sup>3</sup>?

(A)  $2.21 \times 10^{-20}$  g      (B)  $1.63 \times 10^{-20}$  g      (C)  $6.52 \times 10^{-21}$  g      (D)  $1.53 \times 10^{-22}$  g

**PQ-13.** Which sample has the largest volume?

Densities	
aluminum: 2.7 g·cm <sup>-3</sup>	copper: 9.0 g·cm <sup>-3</sup>
(A) 1.0 g aluminum	(B) 5.0 g aluminum
(C) 1.0 g copper	(D) 5.0 g copper

## Chapter 2

**SQ-3.** Which compound is named correctly?

(A) FeCO<sub>3</sub>, iron carbonate      (B) K<sub>2</sub>SO<sub>4</sub>, potassium sulfate  
 (C) Sr(NO<sub>3</sub>)<sub>2</sub>, strontium nitrite      (D) Co<sub>2</sub>(SO<sub>4</sub>)<sub>3</sub>, cobalt(II) sulfate

# Acs Study Guide Answer Key

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Prather Hicks**

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