

1. According to the Brønsted-Lowry definition, which of the following species can function both as an acid and as a base?

- a. CH_4
- b. HCO_3^-
- c. CH_3^+
- d. H_3CO_2^-
- e. H_3O^+

2. In the reaction $\text{Ca(OH)}_2 + \text{H}_2\text{SO}_4 \rightarrow \text{CaSO}_4 + \text{H}_2\text{O}$, which are the acids-base conjugate pair?

- a. H_2O and H_2O^-
- b. H_2O and H_2O^+
- c. Ca^{2+} and H_2O
- d. H_2SO_4 and Ca(OH)_2

3. Given that HNO_3 is a strong Brønsted acid, HNO_2 is a weak acid solution, which is true of a 1 M solution of NaNO_2 ?

- a. It is more basic than a 1 M solution of NaOH .
- b. It is more basic than a 1 M solution of Na_2W .
- c. It yields a neutral solution.
- d. It is more concentrated than a 1 M solution of Na_2Y .

4. The pH of a 0.01 M HCl solution is

- a. 1.5
- b. 2.0
- c. 3.0
- d. 5.0

5. The pH of a 1.0×10^{-3} M Ba(OH)_2 solution at 25°C is

- a. 2.0
- b. 2.5
- c. 11.0
- d. 11.5

6. Which is the strongest acid?

- a. HClO_4
- b. HClO_3
- c. HClO_2
- d. HClO_1

7. Which salt reacts with water (hydration) to generate a basic solution?

- a. $\text{Na}_2\text{CO}_3\text{H}_2\text{O}$
- b. Na_2CO_3
- c. Na_2LiCl
- d. $\text{Na}_2\text{CO}_3\text{O}_2$

8. The position of the first ionization is

Acid	Conjugate Base	K_a (Strength of Acid)
HClO_4	ClO_4^-	0.0001% ionized
HClO_3	ClO_3^-	1.2×10^{-7}
HClO_2	ClO_2^-	2.2×10^{-11}
HClO	ClO^-	1.0×10^{-8}

- a. HClO_4
- b. HClO_3^-
- c. ClO_3^-
- d. ClO_4^-

Acid	Conjugate Base	K_a (Strength of Acid)
H_2O	H_3O^+	1.0×10^{-14}
H_2S	HS^-	1.0×10^{-7}
H_2O_2	HO_2^-	1.0×10^{-14}
H_2O_4^-	HO_3^-	1.0×10^{-14}

9. Which statement is a logical extension from the fact that a 0.1 M solution of potassium acetate KCH_3COO_2 is less acidic than a 0.1 M solution of potassium cyanide, KCN ?

- a. Ethylacetate will be a weaker acid than acetic acid.
- b. Ethylacetate will be less soluble in water than acetic acid.
- c. Cyanide ions are more mobile than acetate ions.
- d. Acetate ions are a weaker acid than hydrocyanic acid.

10. In the situation, will this salt, and its H_2O ionization, act as a strong acid, weak acid, and H_2O salt (neutral), like potassium and chloride ions, or like the conjugate acid with conventional strength, like potassium nitrate?

- a. a 0.1 M sodium sulfide.
- b. a 0.1 M sodium hydroxide solution.
- c. a 0.0001 M potassium nitrate solution.
- d. a 0.0001 M sodium bromate solution.

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The Challenge of Chemistry Philip Archibald Horrigan,1970

Plant Microbiome: Interactions, Mechanisms of Action, and Applications, volume III

Alok Kumar Srivastava,George Newcombe,Gustavo Santoyo ,Prem Lal Kashyap,2025-11-11 Due to the success of the original Research Topic and the following Volume II we have decided to launch a Volume III of Plant Microbiome Interactions Mechanisms of Action and Applications The relationship between plants and their surroundings especially those plant microbe interactions with a beneficial output remain the center of considerable attention Plants depend on their microbiomes for multifarious life supporting activities including nutrient acquisition and augmentation of their defense systems against biotic and abiotic stresses The interactions between a plant and its microbiome are highly complex and dynamic in nature The plant immune system in particular is thought to have a profound impact in determining the plant microbiome structure and diversity Characterizing the plant microbiome and its function could be applied for better crop design and management especially in resource limited environments and to protect plants from intruding pathogens Therefore a deep understanding of the interrelationships of the soil plant microorganism system is essential for improving the efficacy and potential applications of plant growth promoting inocula for achieving sustainable food security and development Recent studies of plant growth promoting rhizobacteria show that cocktails of useful bacteria in the form of synthetic communities are better than single inoculants which face competition from other microorganisms and could be killed or suppressed under sub optimal conditions Microbial activity should also be considered in the construction of microbial co inoculants Understanding plant microbe interactions and mechanisms will also help in the development of smart microbial fertilizer formulation and application in the future This Research Topic welcomes the following subtopics

Interactomics for sustainable agriculture studies using novel combinatorial approaches employing computational expertise to analyze the data from genomics transcriptomics metabolomics and proteomics based studies leading to identification of targets and validating wet lab experiments with molecular evidence for resolving issues in agriculture and sustainability Please note that purely descriptive sequencing studies will not be considered Plant microbiome studies deciphering the structure and composition of complex microbial communities associated directly or indirectly with the plant i e endophytic epiphytic and rhizospheric communities and their significance Topic Editors Alok Kumar Srivastava and Prem Lal Kashyap have created commercial products related to the Topic All other topic editors declare no competing interests with regard to the Research Topic subject

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