

1. According to the Brønsted-Lowry definition, which chemical species can function both as an acid and as a base?

- a.  $\text{Cl}^-$
- b.  $\text{HCO}_3^-$
- c.  $\text{NH}_4^+$
- d.  $\text{H}_2\text{CO}_3$
- e.  $\text{H}_2\text{O}$

2. In the reaction  $\text{CN}^- + \text{H}_2\text{O} \rightleftharpoons \text{HCN} + \text{OH}^-$  which is an acid-base conjugate pair?

- a.  $\text{H}_2\text{O}$  and  $\text{HCN}$
- b.  $\text{H}_2\text{O}$  and  $\text{OH}^-$
- c.  $\text{CN}^-$  and  $\text{H}_2\text{O}$
- d.  $\text{HCN}$  and  $\text{OH}^-$

3. Given that HCN is a stronger Brønsted acid than HF in aqueous solution, which is true of a 1 M solution of NaCN?

- a. It is less basic than a 1 M solution of NaF.
- b. It is more basic than a 1 M solution of NaF.
- c. It yields a neutral solution.
- d. It is more concentrated than a 1 M solution of NaF.

4. The pH of a 0.01 M HCl solution is

- a. 1.5
- b. 2.0
- c. 2.5
- d. 12.5

5. The pH of a  $1.0 \times 10^{-2}$  M  $\text{Na}_2\text{CO}_3$  solution at 25°C is

- a. 2.3
- b. 3.8
- c. 11.0
- d. 11.5

6. Which is the strongest acid?

- a.  $\text{HClO}$
- b.  $\text{HClO}_2$
- c.  $\text{HClO}_3$
- d.  $\text{HClO}_4$

7. Which salt reacts with water (hydrolyzes) to produce a basic solution?

- a.  $\text{Na}_2\text{C}_2\text{H}_3\text{O}_2$
- b.  $\text{Na}_2\text{SO}_4$
- c.  $\text{NH}_4\text{Cl}$
- d.  $\text{BaCl}_2$

8. The weaker of the two acids is

Acid	Conjugate Base	$K_a$ (Acid Ionization Constant of Acid)
$\text{HCl}$	$\text{Cl}^-$	strong acid
$\text{HNO}_3$	$\text{NO}_3^-$	$4.2 \times 10^{-1}$
$\text{H}_2\text{S}$	$\text{HS}^-$	$5.7 \times 10^{-8}$
$\text{H}_2\text{SO}_4$	$\text{HSO}_4^-$	$1.2 \times 10^{-2}$

- a.  $\text{Cl}^-$
- b.  $\text{CN}^-$
- c.  $\text{HS}^-$
- d.  $\text{S}^{2-}$
- e.  $\text{SO}_4^{2-}$

Acid	Conjugate Base	$K_a$ (Acid Ionization Constant of Acid)
$\text{HCl}$	$\text{Cl}^-$	strong acid
$\text{HNO}_3$	$\text{NO}_3^-$	$4.2 \times 10^{-1}$
$\text{H}_2\text{S}$	$\text{HS}^-$	$5.7 \times 10^{-8}$
$\text{H}_2\text{SO}_4$	$\text{HSO}_4^-$	$1.2 \times 10^{-2}$

9. Which statement is a logical inference from the fact that a 0.10 M solution of potassium acetate  $\text{KC}_2\text{H}_3\text{O}_2$  is less alkaline than a 0.01 M solution of potassium cyanide,  $\text{KCN}$ ?

- a. Hydrocyanic acid is a weaker acid than acetic acid.
- b. Hydrocyanic acid is less soluble in water than acetic acid.
- c. Cyanides are less soluble than acetates.
- d. Acetic acid is a weaker acid than hydrocyanic acid.

10. In the titration of 50.0 mL of 0.100 M benzoic acid to monoprotic acid with 25.0 mL of 0.100 M  $\text{NaOH}$ , the properties of the solution at the equivalence point will correspond exactly to the properties of

- a. a 0.100 M sodium solution.
- b. a 0.0500 M sodium hydroxide solution.
- c. a 0.0500 M benzoic acid solution.
- d. a 0.0500 M sodium benzoate solution.

# Acs Chem 1212 Study Guide

**Liyong Dong**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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Due to the success of the original Research Topic and the following Volume II we have decided to launch a Volume III of Plant Microbiome Interactions Mechanisms of Action and Applications The relationship between plants and their surroundings especially those plant microbe interactions with a beneficial output remain the center of considerable attention Plants depend on their microbiomes for multifarious life supporting activities including nutrient acquisition and augmentation of their defense systems against biotic and abiotic stresses The interactions between a plant and its microbiome are highly complex and dynamic in nature The plant immune system in particular is thought to have a profound impact in determining the plant microbiome structure and diversity Characterizing the plant microbiome and its function could be applied for better crop design and management especially in resource limited environments and to protect plants from intruding pathogens Therefore a deep understanding of the interrelationships of the soil plant microorganism system is essential for improving the efficacy and potential applications of plant growth promoting inocula for achieving sustainable food security and development Recent studies of plant growth promoting rhizobacteria show that cocktails of useful bacteria in the form of synthetic communities are better than single inoculants which face competition from other microorganisms and could be killed or suppressed under sub optimal conditions Microbial activity should also be considered in the construction of microbial co inoculants Understanding plant microbe interactions and mechanisms will also help in the development of smart microbial fertilizer formulation and application in the future This Research Topic welcomes the following subtopics Interactomics for sustainable agriculture studies using novel combinatorial approaches employing computational expertise to analyze the data from genomics transcriptomics metabolomics and proteomics based studies leading to identification of targets and validating wet lab experiments with molecular evidence for resolving issues in agriculture and sustainability Please note that purely descriptive sequencing studies will not be considered Plant microbiome studies deciphering the structure and composition of complex microbial communities associated directly or indirectly with the plant i e endophytic epiphytic and rhizospheric communities and their significance Topic Editors Alok Kumar Srivastava and Prem Lal Kashyap have created commercial products related to the Topic All other topic editors declare no competing interests with regard to the Research Topic subject      *Nanotechnology Environmental Health and Safety* Matthew Hull,Diana Bowman,2014-06-11 Nanotechnology Environmental Health and Safety Second Edition focuses not only on the impact of nanotechnology and the discipline of nanotoxicity but also explains each of these disciplines through in the context of management requirements and

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