

9. Which statement is TRUE about the most stable conformation of the molecule?



- (A) the 1-bromo and chloro groups are both equatorial and the bromo group is axial
 (B) the bromo and chloro groups are both equatorial and the 1-bromo group is axial
 (C) the 1-bromo and chloro groups are both equatorial and the chloro group is axial
 (D) all three groups are equatorial

10. Which Newman projection depicts the lowest energy conformation of (R)-3-methylbutane?



11. Which are aromatic?



- (A) I and III
 (B) III and IV
 (C) II and III
 (D) I and II

12. What is the result of hydroboration of 1-methylcyclopentene, followed by oxidation?



- (A) a pair of enantiomers
 (B) a pair of diastereomers
 (C) a meso compound
 (D) two achiral compounds

13. Which molecule will have the smallest HOMO-LUMO energy gap?



14. Which act as a Bronsted base?



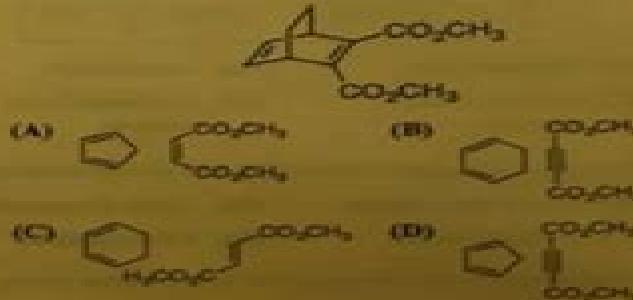
- (A) I
 (B) II
 (C) I and IV
 (D) II and III

15. What is the enantiomeric excess (ee) of the product mixture in this reaction?



- (A) 26% (B) 37% (C) 63% (D) 100%

16. Which reactants should be used to form the product?



17. Which compound will be protonated to the greatest extent in 1.0×10^{-3} M aqueous HCl ?



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**Zayika Wilson-Kennedy,Goldie S.
Byrd,Eugene Kennedy,Henry T.
Frierson**

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Metropolitan Universities ,2006 *Higher Education - Reflections From the Field - Volume 3* ,2023-11-02 COVID wrought havoc on the world s economic systems Higher education did not escape the ravages brought on by the pandemic as institutions of higher education around the world faced major upheavals in their educational delivery systems Some institutions were prepared for the required transition to online learning Most were not Whether prepared or not educators rose to the challenge The innovativeness of educators met the challenges as digital learning replaced the face to face environment In fact some of the distance models proved so engaging that many students no longer desire a return to the face to face model As with all transitions some things were lost while others were gained This book examines practice in the field as institutions struggled to face the worst global pandemic in the last century The book is organized into four sections on

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Green Chemistry Education Paul T. Anastas, Irvin Jay Levy, Kathryn E. Parent, 2009 Green Chemistry has brought about dramatic changes in the teaching of chemistry that have resulted in increased student excitement for the subject of chemistry new lecture materials new laboratory experiments and a world wide community of Green Chemistry teachers This book features the cutting edge of this advance in the teaching of chemistry

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