

25. The number of σ bonds in $N \equiv N$ is
- 1
 - 2
 - 3
 - 4
26. The elements in an ionic compound are held together by
- electrostatic forces of attraction.
 - van der Waals forces
 - the spin of paired electrons.
 - the formation of hybrid orbitals.
 - an electron pair.
27. In every electrolytic and galvanic (voltaic) cell the anode is that electrode
- at which oxidation occurs.
 - which attracts cations.
 - at which electrons are supplied to the solution.
 - at which reduction occurs.
28. Metal X was plated from a solution containing cations of X. The passage of 48.25 C deposited 31 mg of X on the cathode. What is the mass of X (in grams) per mole of electrons?
- 47
 - 62
 - 93
 - 186
29. In a galvanic (voltaic) cell in which the reaction is $Cd + Cu^{2+} \rightarrow Cu + Cd^{2+}$ and the ions are at unit concentration (activity), the cell potential is
- $$\begin{array}{ll} Cd \rightarrow Cd^{2+} + 2e^- & 0.4021 \text{ V} \\ Cu \rightarrow Cu^{2+} + 2e^- & -0.344 \text{ V} \end{array}$$
- 0.1383 V
 - 0.4021 V
 - 0.344 V
 - 0.7461 V
 - 0.3677 V
30. In which reaction will an increase in total pressure at constant temperature favor formation of the products?
- $CaCO_3(s) \rightleftharpoons CaO(s) + CO_2(g)$
 - $H_2(g) + Cl_2(g) \rightleftharpoons 2HCl(g)$
 - $2NO(g) + O_2(g) \rightleftharpoons 2NO_2(g)$
 - $COCl_2(g) \rightleftharpoons CO(g) + Cl_2(g)$

Standard Potentials	E°
$Mg \rightarrow Mg^{2+} + 2e^-$	2.37V
$Al \rightarrow Al^{3+} + 3e^-$	1.66V
$Zn \rightarrow Zn^{2+} + 2e^-$	0.76V
$Fe \rightarrow Fe^{2+} + 2e^-$	0.44V
$Cu \rightarrow Cu^{2+} + 2e^-$	0.34V
$Ag \rightarrow Ag^+ + e^-$	-0.80V

31. Using only the metals Mg, Al, Zn, Fe, Cu and Ag, together with their 1 M salt solutions, a voltaic cell of the highest possible voltage would be constructed using electrodes of these metals.
- Mg and Ag
 - Mg and Fe
 - Zn and Cu
 - Al and Ag
 - Mg and Al
32. $E = E^\circ - 0.059/n \log Q$ (Nernst equation)
 $[H^+] = 1.0 \text{ M}$ initially, $P_{O_2} = 1.0 \text{ atm}$
- $$4e^- + O_2(g) + 4H^+(aq) \rightleftharpoons 2H_2O(l) \quad E^\circ = 1.23 \text{ V}$$
- Based on the information above, which statement is correct?
- $n = 1$, since one mole of oxygen is being considered.
 - Addition of base should result in an E value, which is less than 1.23 V.
 - E is independent of the pH of the solution.
 - $Q = \frac{[H_2O]^2}{[O_2][H^+]}$
33. The equilibrium constant for the gaseous reaction $C + D \rightleftharpoons E + 2F$ is 3.0 at 50 °C. In a 2.0 L flask at 50 °C are placed 1.0 mol of C, 1.0 mol of D, 1.0 mol of E, and 3.0 mol of F. Initially, the reaction will
- proceed at equal rates in both directions.
 - proceed more rapidly to form E and F.
 - proceed more rapidly to form C and D.
 - not occur in either direction.
- | Compound | ΔG_f° , kJ/mol |
|-----------|-----------------------------|
| $H_2O(l)$ | -237 |
| $H_2O(g)$ | -229 |
34. At 298 K the equilibrium constant for $H_2(g) + \frac{1}{2} O_2(g) \rightleftharpoons H_2O(l)$
- is larger than the K_{eq} for $H_2(g) + \frac{1}{2} O_2(g) \rightleftharpoons H_2O(g)$
 - will have a value of 1.0 at equilibrium.
 - cannot be computed since data on O_2 and

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LUCY T. EUBANKS



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Diversity, Equity, and Inclusion for Mathematics and Science Education: Cases and Perspectives Lin, Cheng-Yao, Sun, Li, 2025-06-04 Diversity equity and inclusion DEI are critical pillars for transforming mathematics and science education As classrooms diversify the need to address systemic barriers and create inclusive learning environments becomes more urgent Cases on DEI in STEM education highlight the real world challenges and strategies educators face in promoting equitable access to learning opportunities dismantling biases and empowering students from historically marginalized communities Further exploration may reveal powerful teaching tools and catalyze reflective practice and institutional change encouraging educators to critically examine their roles in shaping a more inclusive future in math and science Cases on Diversity Equity and Inclusion for Mathematics and Science Education explores key issues and concepts related to diversity equity and inclusion in mathematics and science classrooms It offers solutions and successful strategies for teaching and learning in mathematics and science This book covers topics such as inclusive classrooms K 12 education pre service teaching and is a useful resource for educators sociologists academicians researchers and scientists *Acs General Chemistry Study Guide - Acs Exam Prep Secrets, Full-Length Practice Test, Detailed Answer Explanations: [Includes Step-By-Step Video Tutor* Matthew Bowling, 2023-07-11 Mometrix Test Preparation s ACS General Chemistry Study Guide ACS Exam Prep Secrets is the ideal prep solution for anyone who wants to pass their ACS General Chemistry Exam The exam is extremely challenging and thorough test preparation is essential for success Our study guide includes Practice test questions with detailed answer explanations Step by step video tutorials to help you master difficult concepts Tips and strategies to help you get your best test performance A complete review of all general chemistry test sections Mometrix Test Preparation is not affiliated with or endorsed by any official testing organization All organizational and test names are trademarks of their respective owners The Mometrix guide is filled with the critical information you will need in order to do well on your general chemistry exam the concepts procedures principles and vocabulary that the American Chemical Society ACS Examinations Institute expects you to have mastered before sitting for your exam Test sections include Atoms Properties of Matter Bonding and Intermolecular Interactions Reactions Kinetics and Equilibrium Acids and Bases Thermodynamics Electrochemistry Nuclear Chemistry Safety Math and Data in the Laboratory and much more Our guide is full of specific and detailed information that will be key to passing your exam Concepts and principles aren t simply named or described in passing but are explained in detail The Mometrix general chemistry study guide is laid out in a logical and organized fashion so that one section naturally flows from the one preceding it Because it s written with an eye for both technical accuracy and accessibility you will not have to worry about getting lost in dense academic language Any test prep guide is only as good as its practice questions and answer explanations and that s another area where our guide stands out The Mometrix test prep team has provided plenty of general chemistry practice test questions to prepare you for what to expect on the actual exam

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Broadening Participation in STEM Zayika Wilson-Kennedy, Goldie S. Byrd, Eugene Kennedy, Henry T. Frierson, 2019-02-28 This book reports on high impact educational practices and programs that have been demonstrated to be effective at broadening the participation of underrepresented groups in the STEM disciplines

Making it relevant Peter Nentwig, David Waddington, 2006 Teaching in context has become an accepted and often welcomed way of teaching science in both primary and secondary schools The conference organised by IPN and the University of York Science Education Group Context based science curricula drew on the experience of over 40 science educators and 10 projects The book is arranged in four parts Part A consists of two papers one on situated learning and the other on implementation of new curricula Part B contains descriptions of five major curricula in different countries why they were introduced how they were developed and implemented and evaluation results Part C gives descriptions of three projects that are of smaller scale and their materials are used as interventions in other more conventional curricula There is also a contribution on some fundamental research where modules of work are written to examine how best to design context based curricula Finally Part D consist of two chapters one summarising some of the findings that came out of the chapters in the three earlier parts and the second looks at the future

Survival Handbook for the New Chemistry Instructor Diane M. Bunce, Cinzia M. Muzzi, 2004 This book provides an overview of the issues facing new chemistry faculty in preparation for teaching Serving as a reference to answer specific questions new chemistry faculty encounter this book is comparable to sitting down with a colleague in the department and talking through some ideas or gaining some pointers on how to avoid common pitfalls It is the one single place new chemistry faculty can go to find practical information on how to teach and how to prepare for teaching their first course Chapters are written both by established experts in the field and by new professors within their first couple of years of teaching

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