

Name:

Block:

Solutions, Acids, and Bases Test Review

Solutions

- 1- In soda pop, $\text{CO}_2(g)$ is a **solute** and water is the **solvent**.
- 2- What term is used by chemists to quantitatively describe a solution in which a relatively small amount of solute is dissolved? **Dilute/unsaturated**
- 3- When a solvent has dissolved all the solute it can at a particular temperature, it is said to be **Saturated**.
- 4- What will increase the rate of dissolution of a solute in a solvent? **Stirring, increasing surface area, increasing temperature**
- 5- What is the mass of H_2SO_4 in 1.00×10^2 mL of 0.200 M H_2SO_4 solution? **1.96g**
- 6- What number of moles of solute are present in 25.0 mL of 2.00 M HCl ? **0.05 Moles**
- 7- How many grams of K_2SO_4 (molar mass = 174 g/mol) would be needed to prepare 4.00 L of a 0.0510 M solution? **35.5g**
- 8- A solution is prepared by dissolving 7.31 g of Na_2SO_4 in enough water to make 225 mL of solution. Calculate the solution molarity. **0.229M**
- 9- A 54.8 g sample of SrCl_2 is dissolved in 112.5 mL of solution. Calculate the molarity of this solution. **3.07M**

Gas Laws

- 1- A balloon has a volume of 1.20 L at 24.0°C . The balloon is heated to 48.0°C . Calculate the new volume of the balloon. **1.340L**
- 2- A sample of helium gas occupies 2.65 L at 1.20 atm. What pressure would this sample of gas exert in a 1.50 L container at the same temperature? **2.12atm**
- 3- A gas has a volume of 5.0 L at a certain pressure. How must the pressure be changed to double the volume of the gas at constant temperature? **Pressure is 1/2**
- 4- A 6.5 L sample of nitrogen at 25°C and 1.5 atm is allowed to expand to 13.0 L. The temperature remains constant. What is the final pressure? **0.75atm**
- 5- You transfer a sample of a gas at 17°C from a volume of 5.67 L and 1.10 atm to a container at 37°C that has a pressure of 1.10 atm. What is the new volume of the gas? **6.06L**
- 6- A sample of helium gas occupies 12.4 L at 23°C and 0.956 atm. What volume will it occupy at 40°C and 0.956 atm? **13.1L**
- 7- A sample of oxygen gas (O_2) has a volume of 7.51 L at a temperature of 19°C and a pressure of 1.38 atm. Calculate the moles of O_2 molecules present in this gas sample. **0.433 moles**
- 8- A weather balloon contains helium gas at a volume of 855 L. Initially, the balloon is at sea level, where the temperature is 25°C and the barometric pressure is 0.947 atm. The balloon then rises to an altitude of 6000 ft, where the pressure is 0.796 atm and the temperature is 15°C . What is the change in the volume of the balloon as it ascends from sea level to 6000 ft? **128L**
- 9- A 6.35 L sample of carbon monoxide is collected at 55°C and 0.892 atm. What volume will the gas occupy at 1.05 atm and 20°C ? **4.82L**

Acids And Bases Test Review Anwers

Ruth E. McCall

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Iron & Coal Trades Review ,1920 Vol 115 includes Diamond jubilee issue 1867 1927 **Pharmaceutical Record and Weekly Market Review** P. W. Bedford,1892 **Question and Answer Review for Phlebotomy** Kathleen

Becan-McBride,Diana Garza,2000-11 Prentice Hall Health Review Series About the book Q A Review for Phlebotomy by Drs Kathleen Becan McBride and Diana Garza has helped thousands of students pass their certification exams throughout the years Now in its landmark fifth edition the book blends its comprehensive collection of practice exam questions with the exciting possibilities of computer technology As you build confidence by digging into this rich content review you ll find that the Prentice Hall Health test preparation system is a blueprint for success across the boards Boost Your Scores Launch Your Career Our technology solutions allow you to gain more practice and review Log onto www prenhall com review or load the CD ROM that accompanies this book to find additional questions rationales and links to related resources The CD ROM includes an audio glossary and over 750 exam style questions About the Authors Kathleen Becan McBride EdD MT ASCP CLS NCA is the Director of Community Outreach Education at the University of Texas Health Science Center at Houston She has published 13 books and over 50 journal articles She has been educating students in the health professions since 1973 Diana Garza EdD MT ASCP CLS NCA is Associate Research Professor and Project Director for Interdisciplinary Health Care Education at Texas Woman s University Institute of Health Sciences in Houston Texas She is also an Adjunct Clinical Associate Professor in the Department of Laboratory Medicine at the University of Texas M D Anderson Cancer Center in Houston She has published nine books numerous journal articles and has been educating students in the health professions since1981 About the Series Success Across the Boards is a study system that revolutionizes exam preparation for a variety of health care professions By integrating interactive technology with authoritative content and realistic sample test questions Q

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