

1. What type of reaction is shown?



A Combustion

B Neutralization

C Nuclear

D Physical

2. According to Table J, which of the following elements will not cause a reaction with an acid?

A Ag

B Al

C Pb

D Sn

3. Select all of the following that WILL react with H_2 .

A Cobalt

B Copper

C Rubidium

D Sodium

4. Which of the following chemical reactions represents an acid-base reaction?



5. What are the correct products for these reactants: $\text{HCl} + \text{NaOH} \rightarrow$



6. What are the correct products for this reaction $\text{HBr} + \text{Mg}(\text{OH})_2 \rightarrow$



Acid Base Reactions Multiple Choice Questions

James T. Shipman, Jerry D. Wilson

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