

NAME: _____

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DATE: _____

1. What type of reaction is shown?



☐ A Combustion

☐ B Neutralization

☐ C Nuclear

☐ D Physical

2. According to Table J, which of the following elements will not cause a reaction with an acid?

☐ A Ag

☐ B Al

☐ C Pb

☐ D Sn

3. Select all of the following that WILL react with H_2

☐ A Cobalt

☐ B Copper

☐ C Rubidium

☐ D Sodium

4. Which of the following chemical reactions represents an acid-base reaction?



5. What are the correct products for these reactants $\text{HCl} + \text{NaOH} \rightarrow$



6. What are the correct products for this reaction $\text{HBr} + \text{Mg}(\text{OH})_2 \rightarrow$



Acid Base Reactions Multiple Choice Questions

James T. Shipman, Jerry D. Wilson



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