

Answers

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TASK 1 – Bronsted-Lowry acids & bases

- | | | | | | |
|---|---|---|--|---|--|
| 1 | Acid = H_2O , base = NH_3 | 2 | Acid = HCl , base = H_2O | 3 | Acid = HCOOH , base = KOH |
| 4 | Acid = HCl , base = CH_3COOH | 5 | Acid = HCl , base = NH_3 | 6 | Acid = HCO_3^- , base = OH^- |
| 7 | Acid = H^+ , base = HCO_3^- | 8 | Acid = H_2SO_4 , base = HNO_3 | | |

TASK 2 – pH of strong acids

- | | | | | | | | | |
|---|---|-----------------------|---|--------|---|--------|---|----------------------------|
| 1 | a | 0.70 | b | 1.30 | c | 1.10 | d | -0.30 |
| 2 | a | 2.82×10^{-4} | b | 0.0100 | c | 0.0501 | d | 1.58 mol dm^{-3} |
| 3 | a | 1.48 | b | 0.80 | c | 0.10 | d | 0.62 |
| 4 | a | 0.56 | b | 0.39 | c | 0.10 | d | -0.31 |

1	a)	$\text{pH} = -\log 0.2 = 0.70$	
	b)	$\text{pH} = -\log 0.05 = 1.30$	
	c)	$\text{pH} = -\log 0.06 = 1.22$	
	d)	$\text{pH} = -\log 2.00 = -0.30$	
2	a)	$[\text{H}^+] = 10^{-3.10} = 2.82 \times 10^{-4}$	$[\text{HCl}] = 2.82 \times 10^{-4} \text{ mol dm}^{-3}$
	b)	$[\text{H}^+] = 10^{-1.70} = 0.0200$	$[\text{H}_2\text{SO}_4] = 0.0100 \text{ mol dm}^{-3}$
	c)	$[\text{H}^+] = 10^{-1.70} = 0.0501$	$[\text{HNO}_3] = 0.0501 \text{ mol dm}^{-3}$
	d)	$[\text{H}^+] = 10^{-0.30} = 1.58$	$[\text{H}_2\text{SO}_4] = 1.58 \text{ mol dm}^{-3}$
3	a)	$[\text{HNO}_3] = 0.30 \times \frac{50}{500} = 0.0300$	$\text{pH} = -\log 0.0300 = 1.52$
	b)	$[\text{H}_2\text{SO}_4] = 0.40 \times \frac{250}{500} = 0.20$	$\text{pH} = -\log 0.40 = 0.40$
	c)	$[\text{H}_2\text{SO}_4] = 2.00 \times \frac{250}{500} = 1.00$	$\text{pH} = -\log 1.00 = 0.00$
	d)	$[\text{HCl}] = 1.50 \times \frac{250}{500} = 0.75$	$\text{pH} = -\log 0.75 = 0.12$
4	a)	$[\text{HCl}] = \frac{250}{500} \times 0.5 = 0.250$	$\text{pH} = -\log 0.250 = 0.60$
	b)	$[\text{H}_2\text{SO}_4] = \frac{250}{500} \times 0.400 = 0.200$	$\text{pH} = -\log 0.400 = 0.39$
	c)	$[\text{HNO}_3] = \frac{250}{500} \times 0.700 = 0.350$	$\text{pH} = -\log 0.350 = 0.45$
	d)	$[\text{H}_2\text{SO}_4] = \frac{250}{500} \times 1.00 = 0.50$	$\text{pH} = -\log 1.00 = 0.00$

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David R. Klein, Laurie S. Starkey



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A Text-book of Inorganic Chemistry Arnold Frederik Holleman, 1902

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