

## CHEMISTRY UNIT 1 STUDY GUIDE.

### What is Matter?

- Solid, Liquid, gas
- Anything that takes up space & has mass
- Mass is the amount of matter in an object.

### Types

- Substance- a particular kind of matter – pure- made up of elements and compounds
- Mixture- more than one kind of matter. 2+ elements and compounds

### Properties:

**Physical** – property that can be observed and measure without changing the substances.

**Chemical** – property that can only be observed by changing the type of substance.

**Extensive**- only depends on the amount of matter

Mass, weight, volume, length

**Intensive** – only depends on the type of matter, not the amount used to identify a substance.

Color, Oder, melting/ freezing point.

**Ductility**- the ability to make copper into wires.

### States of Matter:

|        | Definite volume | Definite shape                    | Temperature increase | compressable |
|--------|-----------------|-----------------------------------|----------------------|--------------|
| SOLID  | Yes             | Yes                               | Small expans         | No           |
| LIQUID | Yes             | No<br>Takes up shape of container | Small expans         | No           |
| GAS    | No              | No                                | Large expans         | Yes          |



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**David N. Dean**



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## **Table of Contents Science Review Sheet Unit Freshman**

1. Understanding the eBook Science Review Sheet Unit Freshman
  - The Rise of Digital Reading Science Review Sheet Unit Freshman
  - Advantages of eBooks Over Traditional Books
2. Identifying Science Review Sheet Unit Freshman
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Science Review Sheet Unit Freshman
  - User-Friendly Interface
4. Exploring eBook Recommendations from Science Review Sheet Unit Freshman
  - Personalized Recommendations
  - Science Review Sheet Unit Freshman User Reviews and Ratings
  - Science Review Sheet Unit Freshman and Bestseller Lists
5. Accessing Science Review Sheet Unit Freshman Free and Paid eBooks

- Science Review Sheet Unit Freshman Public Domain eBooks
- Science Review Sheet Unit Freshman eBook Subscription Services
- Science Review Sheet Unit Freshman Budget-Friendly Options
- 6. Navigating Science Review Sheet Unit Freshman eBook Formats
  - ePub, PDF, MOBI, and More
  - Science Review Sheet Unit Freshman Compatibility with Devices
  - Science Review Sheet Unit Freshman Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Science Review Sheet Unit Freshman
  - Highlighting and Note-Taking Science Review Sheet Unit Freshman
  - Interactive Elements Science Review Sheet Unit Freshman
- 8. Staying Engaged with Science Review Sheet Unit Freshman
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Science Review Sheet Unit Freshman
- 9. Balancing eBooks and Physical Books Science Review Sheet Unit Freshman
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Science Review Sheet Unit Freshman
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Science Review Sheet Unit Freshman
  - Setting Reading Goals Science Review Sheet Unit Freshman
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Science Review Sheet Unit Freshman
  - Fact-Checking eBook Content of Science Review Sheet Unit Freshman
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

#### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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