

## Section 1 Foundations

PQ-1. The speed of light is  $3.0 \times 10^8 \text{ nm}\cdot\text{ns}^{-1}$ . What is this speed in  $\text{nm}\cdot\text{ms}^{-1}$ ?

(A)  $3.0 \times 10^{-1} \text{ nm}\cdot\text{ms}^{-1}$       (B)  $3.0 \times 10^2 \text{ nm}\cdot\text{ms}^{-1}$       (C)  $3.0 \times 10^{18} \text{ nm}\cdot\text{ms}^{-1}$       (D)  $3.0 \times 10^{23} \text{ nm}\cdot\text{ms}^{-1}$

PQ-4. What is the correctly reported mass of water based on the data in the table?

(A) 1.2 g      (B) 1.20 g

|                          |           |
|--------------------------|-----------|
| Mass of beaker and water | 29.62 g   |
| Mass of beaker only      | 28.3220 g |

(C) 1.298 g      (D) 1.2980 g

PQ-5. What is the name of  $\text{Ti}_2(\text{PO}_4)_3$ ?

(A) titanium phosphate  
 (B) titanium(III) phosphate  
 (C) titanium(IV) phosphate  
 (D) titanium tetrrophosphate

PQ-6. The formula of strontium hexafluorosilicate is  $\text{SrSiF}_6$ . What is the formula of aluminum hexafluorosilicate?

(A)  $\text{AlSiF}_6$       (B)  $\text{Al}_2(\text{SiF}_6)_3$       (C)  $\text{Al}_3\text{SiF}_6$       (D)  $\text{Al}_6(\text{SiF}_6)_2$

PQ-8. What is the correct name for  $\text{N}_2\text{O}_5$ ?

(A) nitrogen(II) oxide  
 (B) nitrogen(V) oxide  
 (C) dinitrogen oxide  
 (D) dinitrogen pentoxide

PQ-12. If a palladium nanoparticle has a density of  $12.0 \text{ g}\cdot\text{cm}^{-3}$ , what is the mass of a nanoparticle with a volume of  $1.84 \times 10^{-21} \text{ cm}^3$ ?

(A)  $2.21 \times 10^{-23} \text{ g}$       (B)  $1.63 \times 10^{-23} \text{ g}$       (C)  $6.52 \times 10^{-21} \text{ g}$       (D)  $1.53 \times 10^{-22} \text{ g}$

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