

Guided Practice - Acids & Bases**Properties of Acids and Bases****1. Match the following terms with their descriptions:**

a. Autoionized Acid _____ the substance formed when a proton is added to _____ a base

b. Autoionized Base _____ the remaining substance when a proton is taken from an acid

c. Conjugate Acid _____ a substance that produces OH^- in solution

d. Conjugate Base _____ a substance producing H^+ in solution

e. Bronsted-Lowry Acid _____ a proton acceptor

f. Bronsted-Lowry Base _____ a proton donor

2. Which of the following represent conjugate acid-base pairs? Label each species as a conjugate acid or a conjugate base. For those pairs that are NOT conjugates, write the correct conjugate acid or base for each species in the pair, labeling each species correctly.

a. HSO_4^- , SO_4^{2-}

b. HBr , BrO^-

c. H_2PO_4^- , PO_4^{3-}

d. H_2O^{2-} , HO^{2-}

3. Complete the following table regarding acid strength.

Property/Condition	Strong Acid	Weak Acid
Conjugate base		
Species present in solution		
Relative strengths of conjugate base compared to water		
Equations following acids in order from weakest		

Acids And Bases Guided Answer Key

Mima M. Horne

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Pocket Guide to Fluid, Electrolyte, and Acid-base Balance Mima M. Horne, 1993 Study Guide for General Chemistry and College Chemistry, Eighth Editions by Holtzclaw and Robinson Norman E. Griswold, Henry

Fuller Holtzclaw, 1988 The Brass World and Platers Guide, 1920 **Organic Chemistry Study Guide & Problems**

Book Brent L. Iverson, 1998 Companion to Organic chemistry 2nd ed by William H Brown. Provides a detailed section by section overview of the major points covered in the text. All of the problems from the text are reprinted here with detailed stepwise solutions *Student's Guide to Fundamentals of Chemistry, Third Edition, Brescia, Arents, Meislich, Turk, Donald Bracken, 1975*

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