

Name: _____ Date: _____ Block: _____

Practice Acid/Base Quiz

Directions: Answer all questions on this practice quiz. Use this as a resource to study for the assessment on Friday, February 28th in addition to your lecture note guides and unit study guide. Your assessment will include both multiple choice and short answer questions.

1. Which of the following could have a pH of 10?

- a. HCl b. HCH_2COO c. NH_4OH d. HF

2. Which of the following is a neutralization reaction?

- a. $\text{CO}_2 + \text{H}_2\text{O} \rightarrow \text{C}_6\text{H}_{12}\text{O}_6 + \text{O}_2$ c. $\text{C}_2\text{H}_6 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$
b. $\text{Ca}(\text{OH})_2 + \text{HNO}_3 \rightarrow \text{H}_2\text{O} + \text{Ca}(\text{NO}_3)_2$ d. $\text{H}_2\text{O} \rightarrow \text{H}_2 + \text{O}_2$

3. Which of the following is directly responsible for acid rain?

- a. steam vented from a nuclear power plant
b. sulfur dioxide released from a coal-fired power plant
c. mining of coal for a coal-fired power plant
d. processing of uranium for a nuclear power plant

Use the following reaction to answer questions 4 through 8.



4. Which compound represents the acid?

- a. LiOH b. H_2O c. LiF d. HF

5. Which compound represents the salt?

- a. LiOH b. H_2O c. LiF d. HF

6. Which compound represents the base?

- a. LiOH b. H_2O c. LiF d. HF

7. Which compound has the highest number of hydrogen (H^+) ions?

- a. LiOH b. H_2O c. LiF d. HF

8. Which compound has the highest number of hydroxide (OH^-) ions?

- a. LiOH b. H_2O c. LiF d. HF

Acid Base Reactions Multiple Choice Questions

Joseph A. Mascetta



Acid Base Reactions Multiple Choice Questions:

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Medicinal Chemistry Sriram D., 2010-09 The second edition of Medicinal Chemistry is based on the core module of pharmacy syllabi of various technical universities and targets undergraduate B Pharma students across India The current edition has been designed by authors based on the opinion of the experts to include the latest developments in the field of medicinal chemistry detailed synthesis mechanism of the drugs and their mode of action inside the body Chemistry Nicholas R. Romano, Philip K. Cameron, 2002 Prepares students for the new standards and the commencement level PS Chemistry Test Challenges with content based multiple choice constructed response and real world thematic questions Stimulates skills based activities in reading writing and lab performance Correlates PS Chemistry key ideas and performance indicators on atomic concepts periodic table moles stoichiometry bonding behavior of matter kinetics organic chemistry oxidation reduction acids bases and salts nuclear chemistry Fosters mastery with practice on three recent tests **Solving Equilibrium Problems with Applications to Qualitative Analysis** Steven S. Zumdahl, 1989 *5 Steps to a 5 AP Chemistry, 2008-2009 Edition* John T. Moore, Richard H. Langley, 2008-01-04 A PERFECT PLAN FOR THE PERFECT SCORE We want you to succeed on your AP exam That's why we've created this 5 step plan to help you study more effectively use your preparation time wisely and get your best score This easy to follow guide offers you a complete review of your AP course strategies to give you the edge on test day and plenty of practice with AP style test questions You'll sharpen your subject knowledge strengthen your thinking skills and build your test taking confidence with Full length practice exams modeled on the real test All the terms and concepts you need to know to get your best score Your choice of three customized study schedules so you can pick the one that meets your needs The 5 Step Plan helps you get the most out of your study time Step 1 Set Up Your Study Program Step 2 Determine Your Readiness Step 3 Develop the Strategies Step 4 Review the Knowledge Step 5 Build Your Confidence **First Year Chemistry Students' Conceptions of Acid/base Chemistry** Sally Diane Rupert, 2001 **Shipman Phys Sci 6e Study Guide** James T. Shipman, Jerry D. Wilson, 1990 **Tests** Richard C. Sweetland, Daniel J. Keyser, 1983 Descriptions of over 3000 tests in English intended as a guide for psychologists educators and other personnel who need test information to meet their assessment needs Topical arrangement under 3 main sections of psychology education and business Each entry gives test name associated personal names intended age group purpose description time range scoring cost and publisher Indexes by test titles authors publishers visually impaired tests and scoring services **5 Steps to a 5: AP Chemistry 2021** Mary Millhollon, Richard H. Langley, 2020-10-02 MATCHES THE LATEST EXAM In this hybrid year let us supplement your AP classroom experience with this multi platform study guide The immensely popular 5 Steps to a 5 AP Chemistry guide has been updated for the 2020-21 school year and now contains 3 full length practice exams available both in the book and online that reflect the latest exam Up to Date Resources for COVID 19 Exam Disruption Access to a robust online platform Comprehensive overview of the AP Chemistry exam format Hundreds of

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Concepts in Biology Eldon D. Enger, Frederick C. Ross, David B. Bailey, 2005 Concepts in Biology is a short student friendly text organized in a traditional manner It has very little botany and presents a human oriented approach to the animal unit Professors and students appreciate the low cost of this title and that it is written for students who are not biology majors

Physical science Jay M. Pasachoff, LeVon Balzer, Phyllis L. Goodson, Ann Collins, 1983

Chemistry David W. Oxtoby, Wade A. Freeman, Toby F. Block, 1998 Textbook of chemistry concepts

Science Education International, 2005

How to Prepare for the SAT II Joseph A. Mascetta, 1998 An extensive review of chemistry topics a full length diagnostic test precedes four up to date full length model SAT II Chemistry Tests All tests are followed with answers charts students can use to diagnose their individual study needs Chemistry topics reviewed include atomic structure bonding all other topics included in the SAT II test Joseph A Mascetta is a science educational consultant a former principal chemistry teacher at Mount Lebanon High School Pittsburgh PA

Acid Base Reactions Multiple Choice Questions Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has are more apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Acid Base Reactions Multiple Choice Questions**," written by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound affect our existence. Throughout this critique, we shall delve in to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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