

# ACS Biochemistry Final

pKa (chapter 3)

measure of the tendency of a group to give up a proton (acidity); tendency decreases tenfold as pKa increases by one unit

thioester bond (1)

compounds with functional group C-S-CoA-C (eg, in acetyl-CoA)

zwitterion (3)

dipolar ion with both positive and negative groups but overall neutral charge; can act as either an acid or base

Isoelectric point (pI) (chapter 3)

the characteristic pH at which the net electric charge is zero

SDS - Sodium Dodecyl Sulfate (chap 3)

detergent used to unfold proteins and give them uniform negative charge

SDS Page (3)

chromatography used to separate proteins based on mass. light proteins travel fast than heavier ones

PCR - Polymerase chain reaction (3)

copies DNA multiple times to increase sample size

Isoelectric focusing (chapt 3)

procedure used to determine the isoelectric pt (pI) of a protein. Protein migrates through gel until pH = pI (net charge = 0)

Two-Dimensional Electrophoresis (3)

combines isoelectric focusing and SDS electrophoresis; separates proteins by both molecular weight and pI

specific activity (3)

number of enzyme units per mg of total protein (a measure of enzyme purity)

activity (3)

total units of a certain enzyme in a solution

Peptide bonds (chapter 4)

C-N bond with double bond character due to resonance (C-N bond cannot rotate, and is planar)

Edman degradation (3)

used in the sequencing of polypeptides; labels and removes ONLY the amino-residue from a polypeptide. carried out in a machine called a sequenator

$\phi$  in peptide bonding (chapter 4)

angle around the  $\alpha$ -carbon - amide nitrogen bond

$\psi$  in peptide bonding (chapter 4)

angle around the  $\alpha$ -carbon - carbonyl carbon bond

Ramachandran Plot (4)

shows favorable  $\phi$ - $\psi$  angle combinations. 3 main "wells" for  $\alpha$ -helices,  $\beta$ -sheets, and left handed  $\alpha$ -helices

Levinthal's Paradox (4)

# Advanced Biochemistry Acs Exam Study Guide

**Rachel S Tattersall**



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