

## CHEMISTRY UNIT 1 STUDY GUIDE.

### What is Matter?

- Solid, Liquid, gas
- Anything that takes up space & has mass
- Mass is the amount of matter in an object.

### Types

- Substance- a particular kind of matter — pure - made up of elements and compounds
- Mixture- more than one kind of matter. 2+ elements and compounds

### Properties:

Physical – property that can be observed and measured without changing the substance.

Chemical – property that can only be observed by changing the type of substance.

Extensive- only depends on the amount of matter

Mass, weight, volume, length

Intensive – only depends on the type of matter, not the amount used to identify a substance.

Color, Odor, melting/ freezing point.

Ductility- the ability to make copper into wires.

### States of Matter:

	Definite volume	Definite shape	Temperature increase	compressible
SOLID	Yes	Yes	Small expans	No
LIQUID	Yes	No Takes up shape of container	Small expans	No
GAS	No	No	Large expans	Yes



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**University of Chicago**

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