

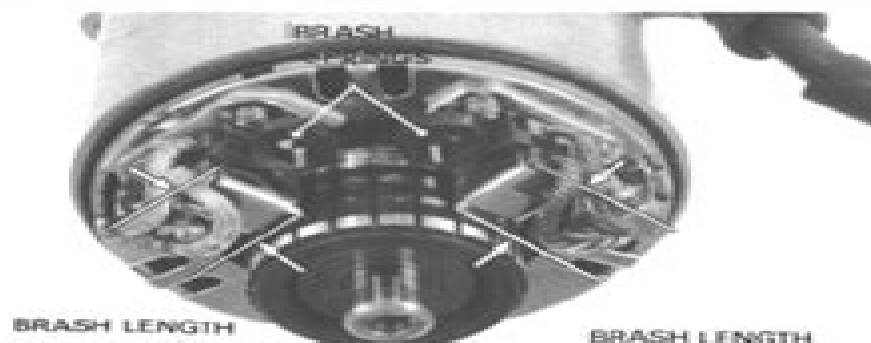


BRUSH INSPECTION

Remove the starter motor case screen.
Inspect the brushes and measure the brush length.
Measure brush spring tension with a spring scale.

SERVICE LIMITS:

Brush length: 7.5 mm (0.30 in.)
Brush spring tension: 560 g (19.75 oz)



COMMUTATOR INSPECTION

Remove the starter motor case.

NOTE

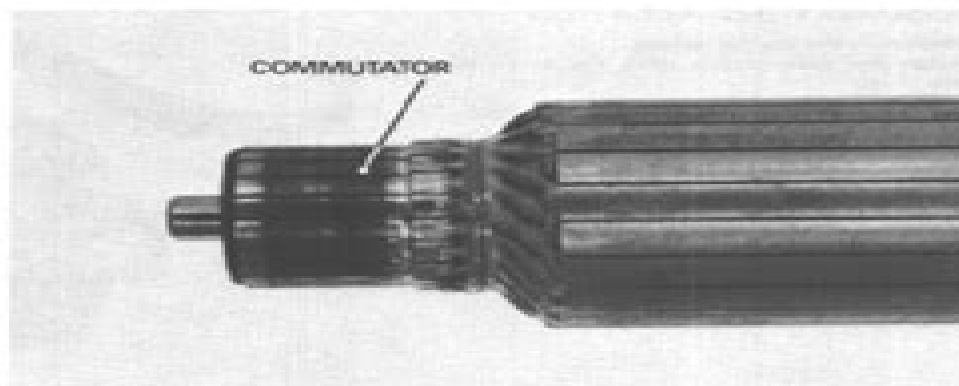
Record the location and number of thrust washers.

Inspect the commutator bars for discoloration.
Bars discolored in pairs indicate grounded armature coils.

NOTE

Do not use emery or sand paper on the commutator.

Check for continuity between pairs of commutator bars, and also between commutator bars and armature shaft.



1982 Honda Nighthawk 750 Service Manual

Pete Shoemark, Ken Freund



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web correct answer c both a and b q2 the density of 1 m solution of hcl is 1 0585 g ml the molality of the solution is 1 0585 1 0 10 0 0585 correct answer b 1 explanation molality begin array l m left frac 1000 m 1000d mm right end array

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web 1 molarity mv mass molar mass $\times 0.4500$ l 0.825 g 141.9579 g/mol $\times 0.0129$ m 2 molality 0.825 g 141.9579 g/mol 0.00581158 mol 0.00581158 mol 0.4500 kg 0.0129 m 3 mole fraction n_2 hpo 4 0.825 g 141.9579 g/mol 0.00581158 mol h₂o 450.0 g 18.015 g/mol 24.97918401 mol

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web apr 7 2014 molarity moles of solute liters of solution 8.422 a first convert 250 ml to liters 250/1000 = 0.25 then calculate molarity 5 moles / 0.25 liters = 20 M 3 c a solution with molarity 2 requires 2 M of NaOH per liter so 4 x 2.8 M = 11.2 M a solution of molarity 1.5 M requires 1.5 mol of NaOH to every litre of solvent

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web answer the correct option is c explanation the molarity of a mixture m_{mix} can be calculated using the following formula $m_{mix} = \frac{m_1 + m_2}{V_1 + V_2} = \frac{1.5 \times 480 + 1.2 \times 520}{480 + 520} = \frac{720 + 624}{1000} = 1.344$ M q5 with increase in temperature which of these changes a molality b fraction of solute present in water

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