



EARTH SCIENCE

INTRODUCTION

COMPONENTS OF STUDY

- Geology: Earth physical & historical
- Metgeology: Weather & Climate
- Oceanography: The oceans
- Astronomy: What is external to the atmosphere

THE FOUR SPHERES (See The Water Cycle)

- Lithosphere: The solid earth
- Hydrosphere: The liquid earth
- Atmosphere: The gaseous earth
- Biosphere: Where life exists

CONCERN IN OUR PHYSICAL ENVIRONMENT

- Resources: Water, soil, metallic & non-metallic elements, sunlight, sunlight, wind, etc.
- The capita use of metallic and non-metallic elements is 11 tons per year
- Renewable vs. non-renewable resources
- Environmental and population problems

MINERALS

- A mineral is a naturally occurring, inorganic, solid material with a definite atomic structure.
- An ore is a useful mineral resource.

A. Atom and Minerals

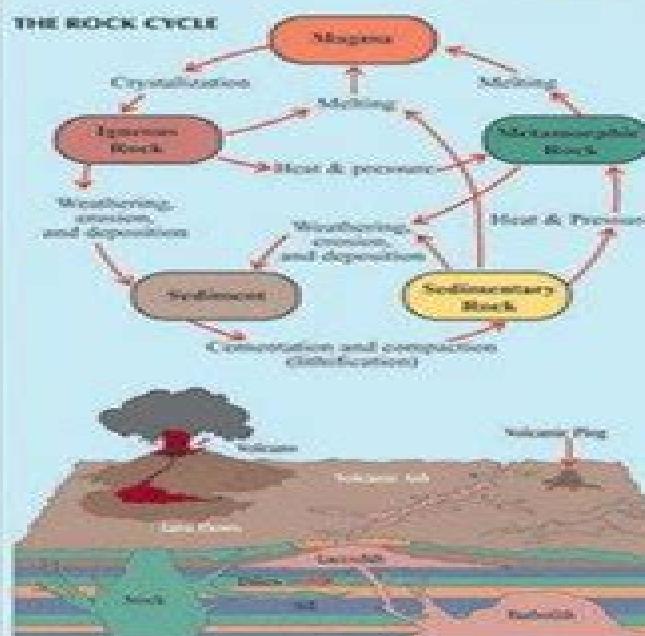
- Atoms in ancient Greek means "un-cut" i.e., the Greeks believed the atom to be the smallest particle of matter.
- Atoms are composed of neutrons, protons and electrons.
- Atomic number: Number of protons in nucleus.
- Atomic weight: Average weight of an atom.
- Isotopes: Atoms with the same atomic number but different numbers of neutrons in the nucleus.
- Forces: Strong, weak, electromagnetic and gravity.
- Properties of Minerals**
 - Color: Appearance or quality of light from surface.
 - Texture: Nature of light and cause of color.
 - Streak: Color of material in powdered form.
 - HARDNESS: Mohs' Scale (See Figure Below).
 - Scratch Test: Higher numbered materials can scratch lower numbered materials.
 - Shape is determined by cleavage, crystal form and structure.
- Specific Gravity: The ratio of the weight of a mineral to the weight of an equal volume of water.
- Density of water = 1 gram/cubic centimeter.
1 g/cm³ = 7.7, aluminum = 2.7, iron = 7.2.

MOM'S SCALE		MINERALS
10	Diamond	
9	Corundum	
8	Topaz	
7	Quartz	
6	feldspar	
5	Spatite	
4	Fluorite	
3	Garnet	
2	Pyrope	
1	Redite	
0	Table	

MINERAL CONCEPTS

- Silicates: Minerals with silicon and oxygen.
- Silicates are a large class which forms a pyramidal-shaped structure with oxygen.
- Other forms include simple non-silicate minerals, oxides, sulfides, noble elements, 2-3 element hydrocarbons and quartz.
- Non-Silicates
- Make up one-fourth of continental crust.
- Carbides: Minerals with carbon and oxygen including calcium, those which are present limestone (calcium) and marble (calcium carbonate).
- Oxides: Oxygen-banded solids, i.e., iron sulfides (FeS), sulfates (MgSO₄), halides (CaCl₂, F).
- Halite: Mineral form of salt.
- Corundum: Plasma, calcium.
- Noble metals: Iron, zinc, gold, silver, nickel.

THE ROCK CYCLE



ROCK TYPES

METAMORPHIC ROCKS

A. Formation

- Metamorphism:** To change form.
- Metamorphic rock-formations:** Material under intense pressure, high temperatures.
- Contact metamorphism:** Metamorphism caused by pressure to change.
- Low-grade metamorphism:** Shallow metamorphic rocks.
- High-grade metamorphism:** Deeper metamorphic rocks.
- foliation:** During metamorphism, parallel mineral lamellae are formed that are oriented in the same direction.
- Metamorphic structures:** Features of material that change with heat with out melting.
- Agents of Metamorphism:**

B. Heat

- most important agent
- possible energy for chemical reactions
- For example, clay becomes shale at increased in great temperature

C. Chemical Activity

- Metamorphic rocks:** Fine-grained, dark, shiny rock with randomly arranged minerals.
- Shale:** Fine-grained, often grey, foliated rock, light grey along cleavage, pieces of rock broken by pressure.
- Marble:** Calcium or magnesium-foliated rock, formed from limestone.
- Phyllite:** Shiny, foliated rock, more compactly foliated than shale.
- Schist:** Foliated rock, more coarsely granular and of higher metamorphic grade than phyllite.
- Gneiss:** Foliated, banded rock of the highest metamorphic grade.
- Quartzite:** Very hard, granular quartz rock, formed from sandstone.

IGNEOUS ROCKS

A. Formation

- Extrusive igneous rocks:** Formed in volcanic eruptions at surface.
- Extrusive or plutonic igneous rocks:** Produced within the earth.
- Magmatic:**
 - Contains mainly of silicate minerals.
 - Has gasses, such as water vapor.
 - Temperature there are about 2000 °K.
 - Causes an rise of cooling, chemical composition, and amount of gases.
 - Produces intrusive igneous rocks.
- Igneous:**
 - Similar to igneous.
 - Most gaseous elements have escaped.
 - Produces extrusive (volcanic) igneous rocks.
- Metamorphic:**
 - Can stabilize different patterns of ions.
 - The rate of cooling strongly affects crystal size.
 - Slow cooling = Large crystals.
 - Quick cooling = Small crystals.
- Amorphous (non-crystalline) pattern of ions:**
 - rocks with intense cooling (quenching).
 - produces rocks referred to as glass.

C. Classifying Igneous Rocks

- By texture:** Size and arrangement of interlocking crystals.
 - Fast-cooled:
 - Form at surface or within upper crust.
 - Crystals left the gas bubbles are called vesicles.
 - Slow-cooled:
 - Formed deep within the interior.
 - Large crystals, embedded in matrix of smaller crystals.
 - By mineral composition:**
 - Depends on composition of magma and cooling.
 - Some magmas can create rocks of varying mineral content.
 - As magma cools, certain minerals crystallize first (e.g. felsic to mafic).

Igneous Rock Type	Igneous Rock		Rock Type
	Extrusive	Plutonic	
High Silica	• Obsidian	• Gabbro	• Igneous
Medium Silica	• Pumice	• Andesite	• Metamorphic
Low Silica	• Basalt	• Diorite	• Sedimentary

Earth Science High School Essential Question

Huangqi Zhang

Earth Science High School Essential Question:

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Science Curriculum Resource Handbook ,1992 **Intersections of Formal and Informal Science** Lucy Avraamidou,Wolff-Michael Roth,2016-03-10 Science learning that takes place between and at the intersections of formal and informal science environments has not been systematically reviewed to offer a comprehensive understanding of the existing knowledge base Bringing together theory and research this volume describes the various ways in which learning science in various settings has been conceptualized as well as empirical evidence to illustrate how science learning in these settings can be supported

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Publisher **Proceedings of the ... Annual Conference** Science Teachers Association of New York State,1904

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